

CFD MODELING OF TRICKLE BED REACTORS: Hydro-processing Reactor

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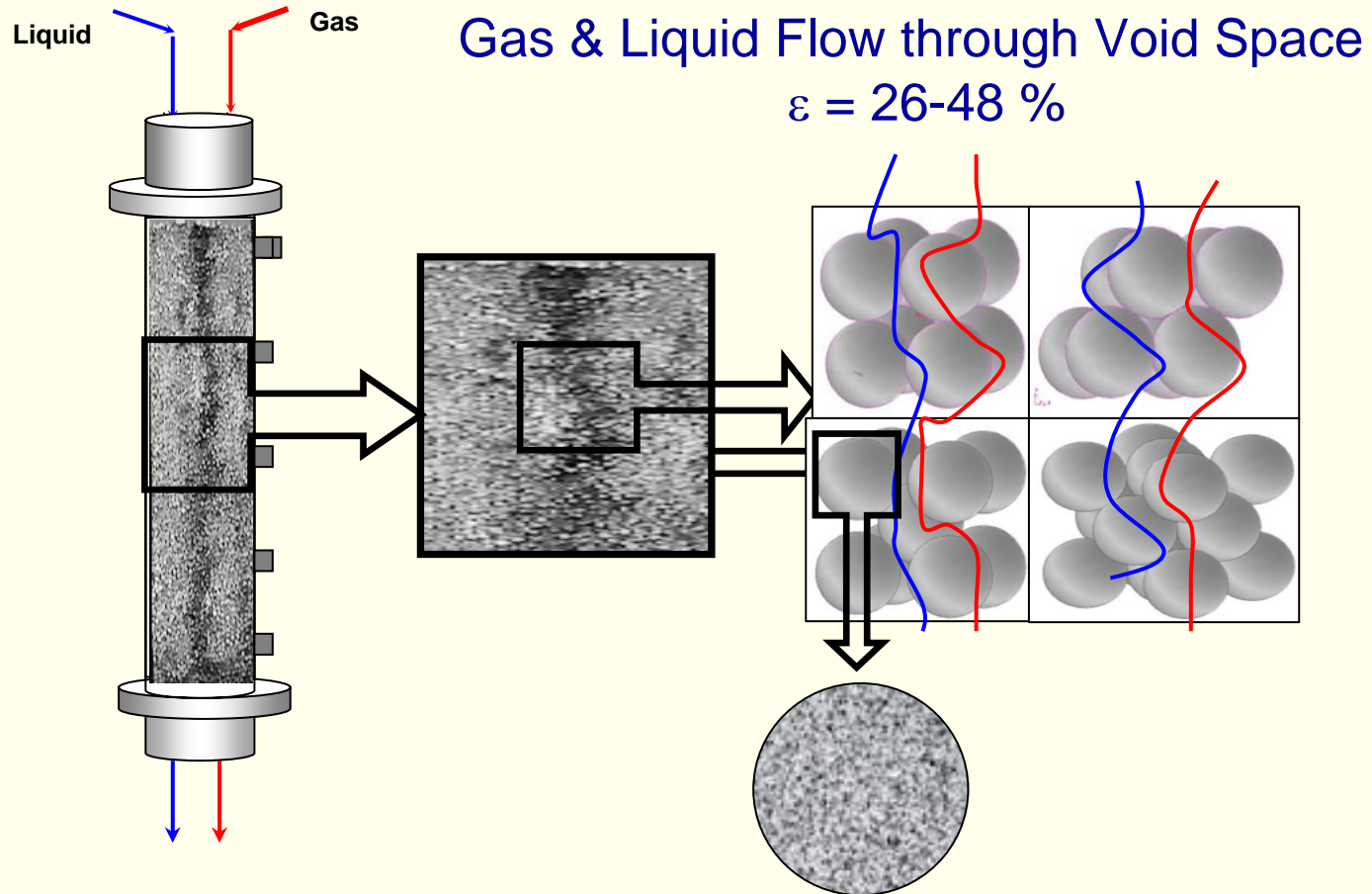


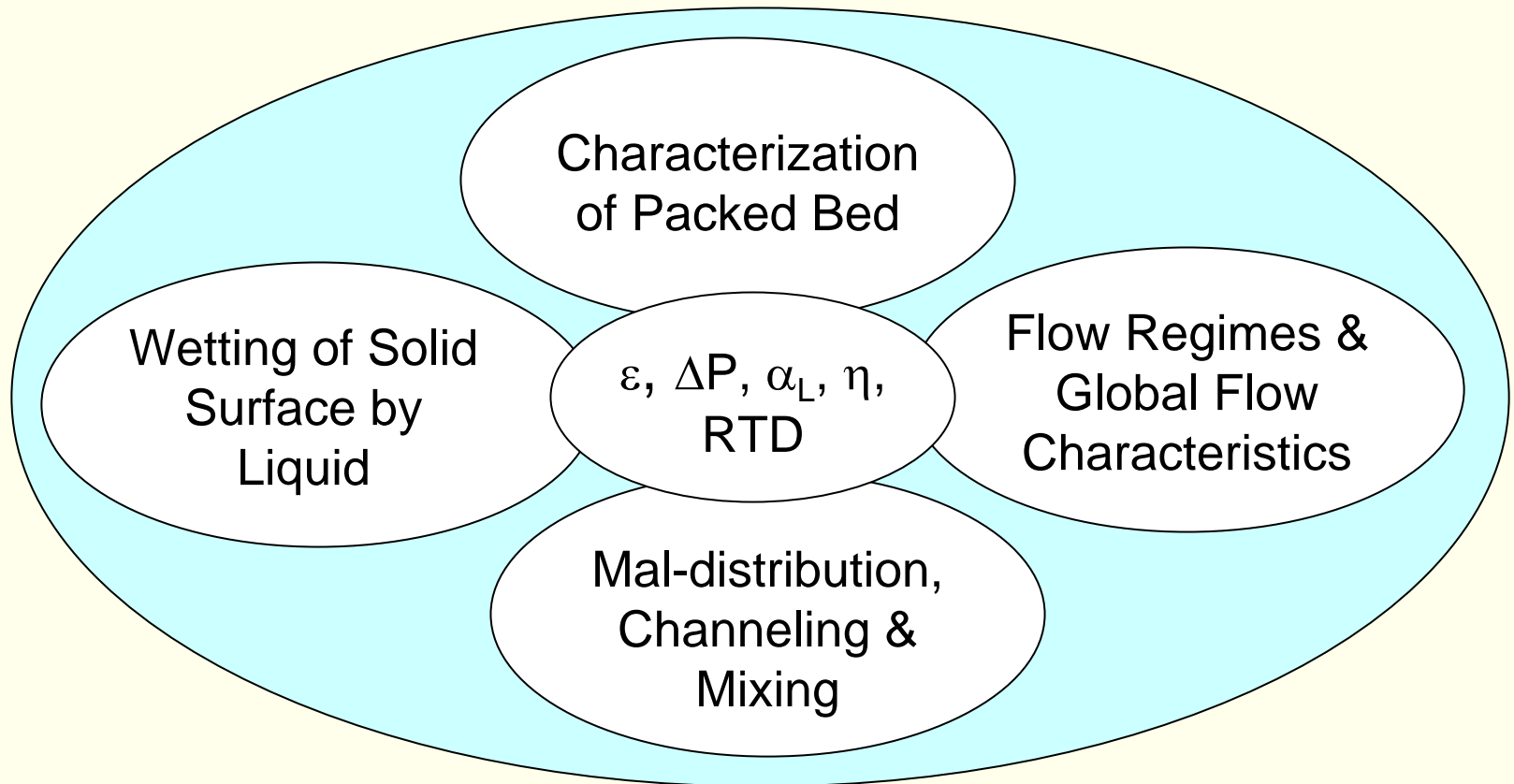
- **Trickle Bed Reactors**
 - Applications/ flow regimes
 - Experiments
- **CFD Modeling of Trickle Bed Reactors**
 - Inter-phase momentum exchange/ capillary terms
 - Estimation of fraction of liquid suspended in gas phase
- **Simulating Hydro-processing Reactor**
 - Reaction kinetics/ other sub-models
 - Influence of reactor scale
- **Concluding Remarks**

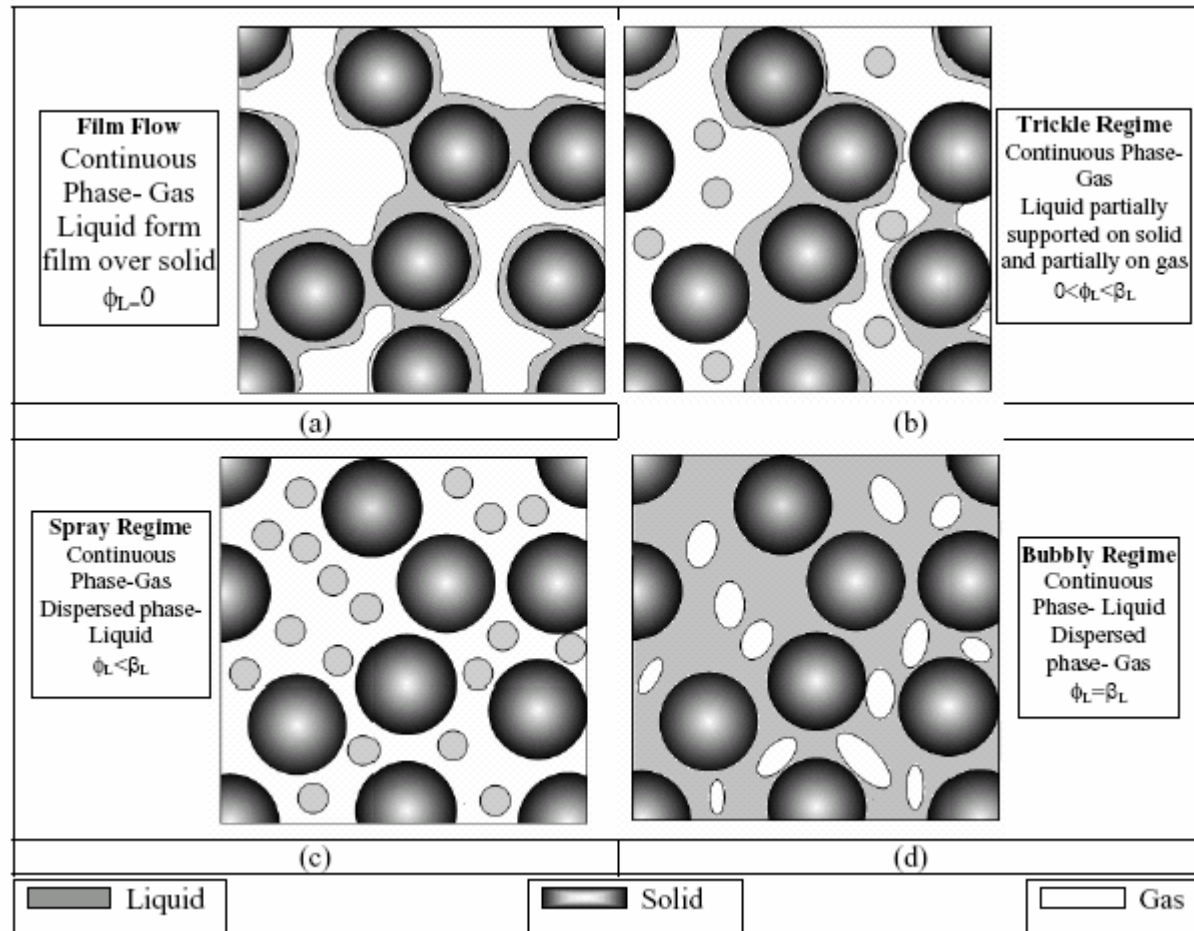


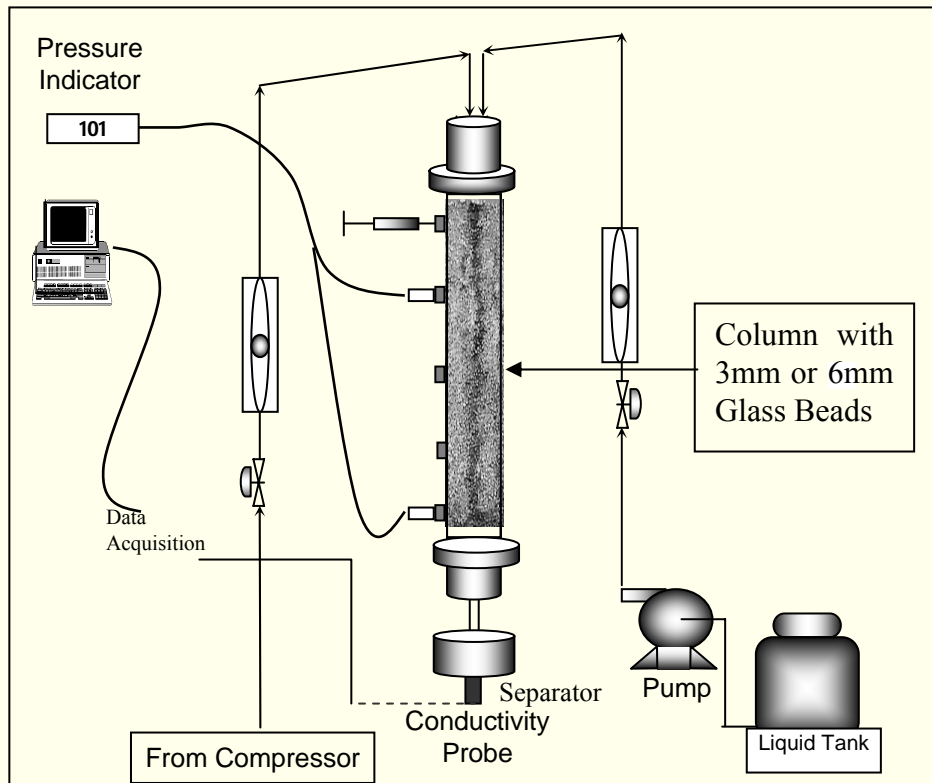
- **Wide Applications**
 - Hydro-de-sulfurization
 - Hydro-cracking/ hydro-treating
 - Hydrogenation/ oxidation
 - Waste water treatment
- **Key Characteristics**
 - Close to plug flow/ Low liquid hold-up
 - Suitable for slow reactions

 - Poor heat transfer/ possibility of mal-distribution
 - Difficult scale-up

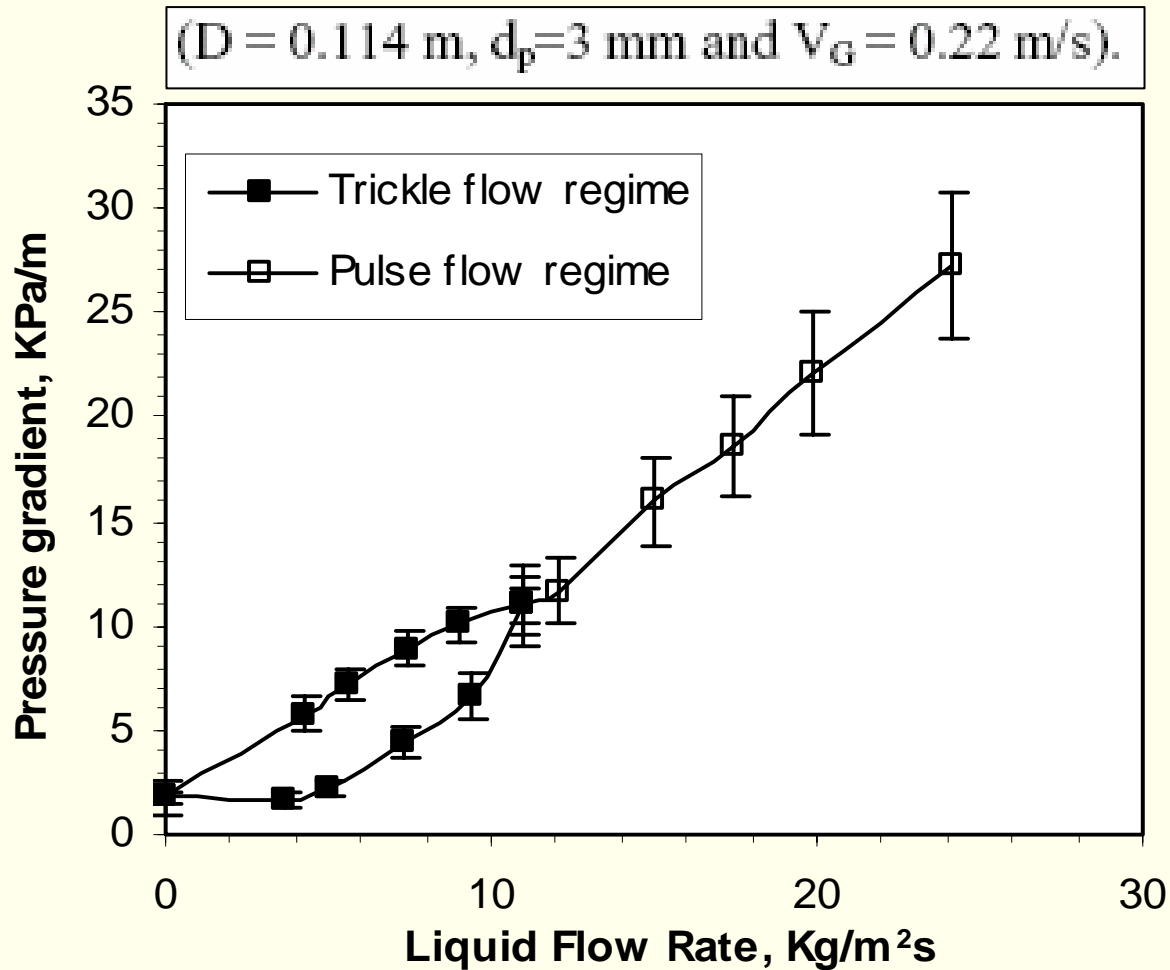


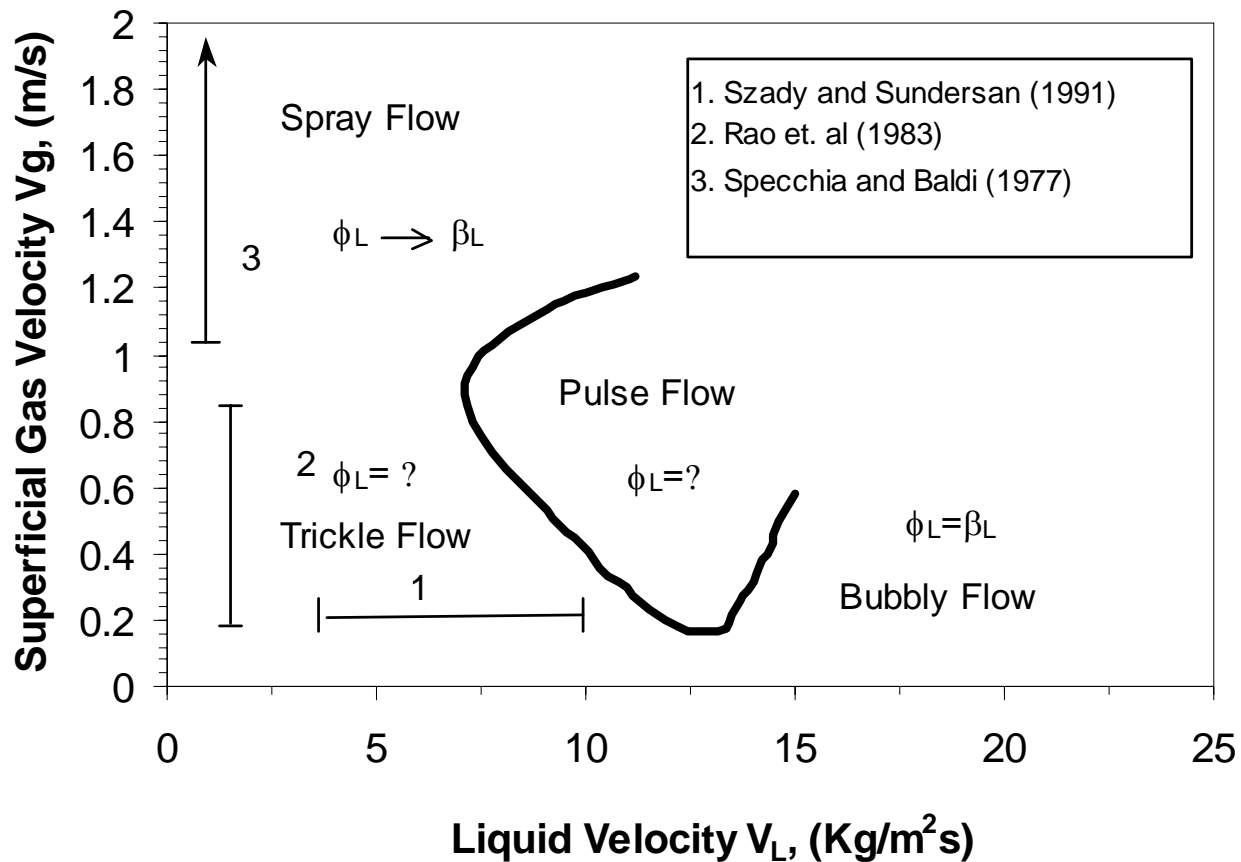






- Hydrodynamics
 - Pressure Drop
 - Liquid Hold-up
- Conductivity Probes
 - Residence Time Distribution
- Liquid Distribution at Inlet
 - Uniform
 - Non-uniform
- Experimental Parameters
 - Bed Diameter: 0.1 & 0.2 m
 - Particle Diameter: 3 & 6 mm
 - Liquid Velocity: < 24 mm/ s
 - Gas Velocity: < 0.50 m/ s
 - Tracer : NaCl







- **Bed Porosity: Scale of Scrutiny**
 - Bed diameter/ length: overall ε
 - Intermediate scale: Gaussian distribution
 - Smaller than particle diameter: Bi-modal distribution
- **Approach Used:**
 - Experimentally measured or estimated radial variation of axially averaged bed porosity
 - Specify porosity by drawing a sample assuming Gaussian distribution with specified variance



- Radial Variation of Porosity: Mueller (1991)

$$\varepsilon(r) = \varepsilon_B + (1 - \varepsilon_B) J_0(ar^*) e^{-br}$$

$$a = 8.243 - \frac{12.98}{(D/d_p - 3.156)} \quad \text{for } 2.61 \leq D/d_p \leq 13.0$$

$$a = 7.383 - \frac{2.932}{(D/d_p - 9.864)} \quad \text{for } 13.0 \leq D/d_p$$

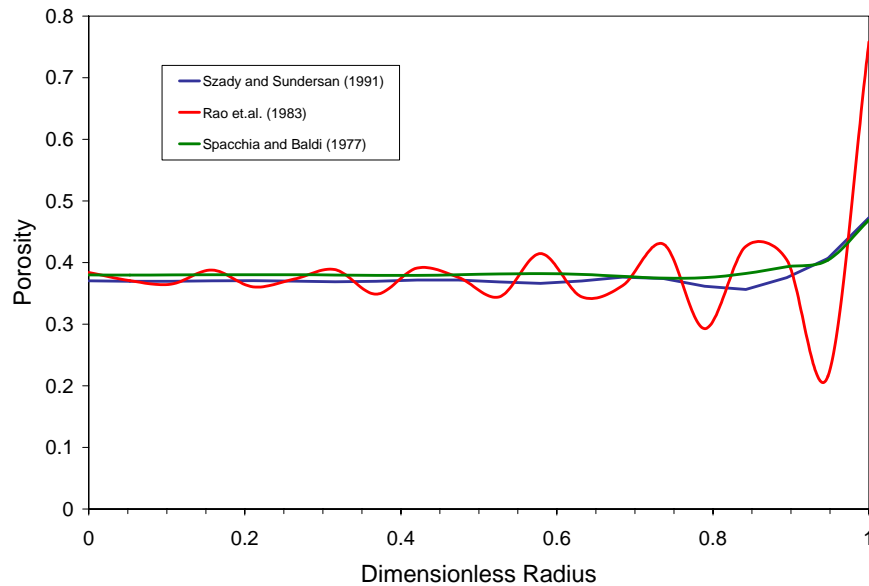
$$b = 0.304 - \frac{0.724}{D/d_p}$$

$r^* = r/D$ and J_0 is zeroth order Bessel Function

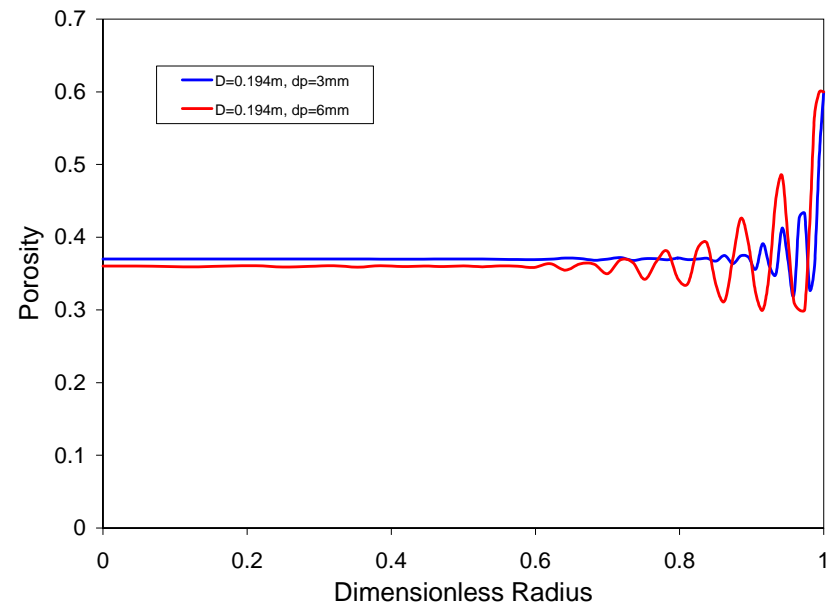


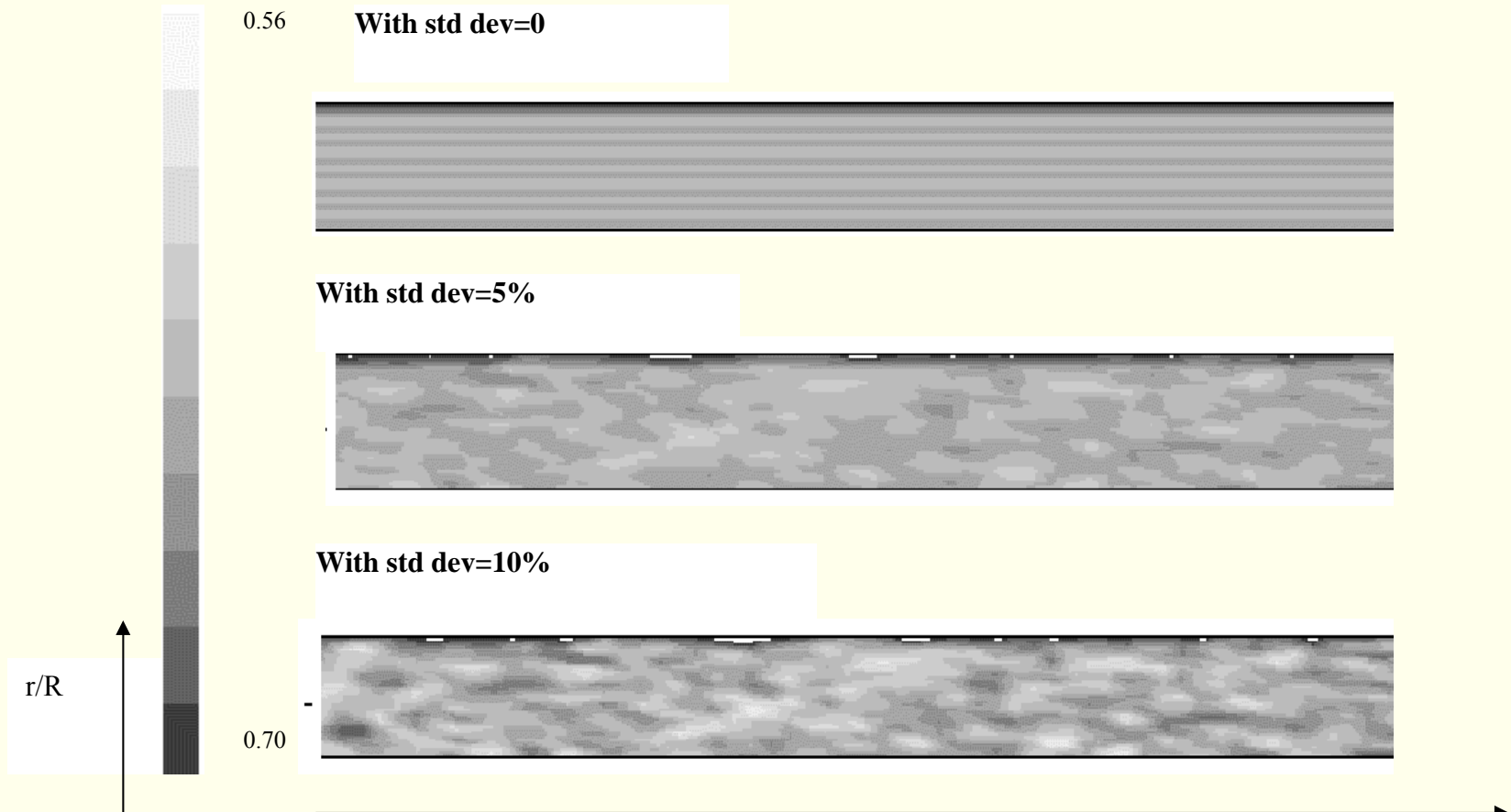
- Mueller's Correlation

Three data sets from literature



Our experiments





- Selection of appropriate value is not straight forward: RTD may help



- Multi-fluid Model (Eulerian-Eulerian)

Continuity Equation

$$\frac{\partial \varepsilon_K \rho_K}{\partial t} + \nabla \cdot \varepsilon_K \rho_K U_K = 0$$

Momentum Balance Equation

$$\frac{\partial (\varepsilon_K \rho_K U_K)}{\partial t} + \nabla \cdot (\varepsilon_K \rho_K U_K U_K) = -\varepsilon_K \nabla P + \nabla \cdot (\varepsilon_K \mu \nabla U) + \varepsilon_K \rho_K g + F_{K,R} (U_k - U_R)$$



- Inter-phase Coupling Terms (Attou & Ferschneider, 2000)

$$F_{GL} = \varepsilon_G \left(\frac{E_1 \mu (1 - \varepsilon_G)^2}{\varepsilon_G^2 d_p^2} \left[\frac{\varepsilon_L}{(1 - \varepsilon_G)} \right]^{0.667} + \frac{E_2 \rho_P (U_G - U_L) (1 - \varepsilon_G)}{\varepsilon_G d_p} \left[\frac{\varepsilon_L}{(1 - \varepsilon_G)} \right]^{0.333} \right)$$

$$F_{GS} = \varepsilon_G \left(\frac{E_1 \mu (1 - \varepsilon_G)^2}{\varepsilon_G^2 d_p^2} \left[\frac{\varepsilon_S}{(1 - \varepsilon_G)} \right]^{0.667} + \frac{E_2 \rho_P U_G (1 - \varepsilon_G)}{\varepsilon_G d_p} \left[\frac{\varepsilon_S}{(1 - \varepsilon_G)} \right]^{0.333} \right)$$

$$F_{LS} = \varepsilon_L \left(\frac{E_1 \mu \varepsilon_s^2}{\varepsilon_L^2 d_p^2} + \frac{E_2 \rho_P U_G \varepsilon_S}{\varepsilon_L d_p} \right)$$



- Capillary Terms

$$P_G - P_L = 2\sigma \left(\frac{1}{d_1} - \frac{1}{d_2} \right)$$

- Attou and Ferschneider (2000)

$$P_G - P_L = 2\sigma \left(\frac{1 - \varepsilon}{1 - \varepsilon_G} \right)^{0.333} \left(\frac{0.5416}{d_p} \right) F \left(\frac{\rho_G}{\rho_L} \right)$$

$$\frac{\partial P_G}{\partial z} - \frac{\partial P_L}{\partial z} = \frac{2}{3} \sigma \frac{5.416}{d_p} \left(\frac{\varepsilon_s}{1 - \varepsilon_G} \right)^{-2/3} \left(\left(\frac{1}{1 - \varepsilon_G} \right) \frac{\partial \varepsilon_s}{\partial z} + \left(\frac{\varepsilon_s}{(1 - \varepsilon_G)^2} \right) \frac{\partial \varepsilon_G}{\partial z} \right) F \left(\frac{\rho_G}{\rho_L} \right)$$



- Capillary Terms

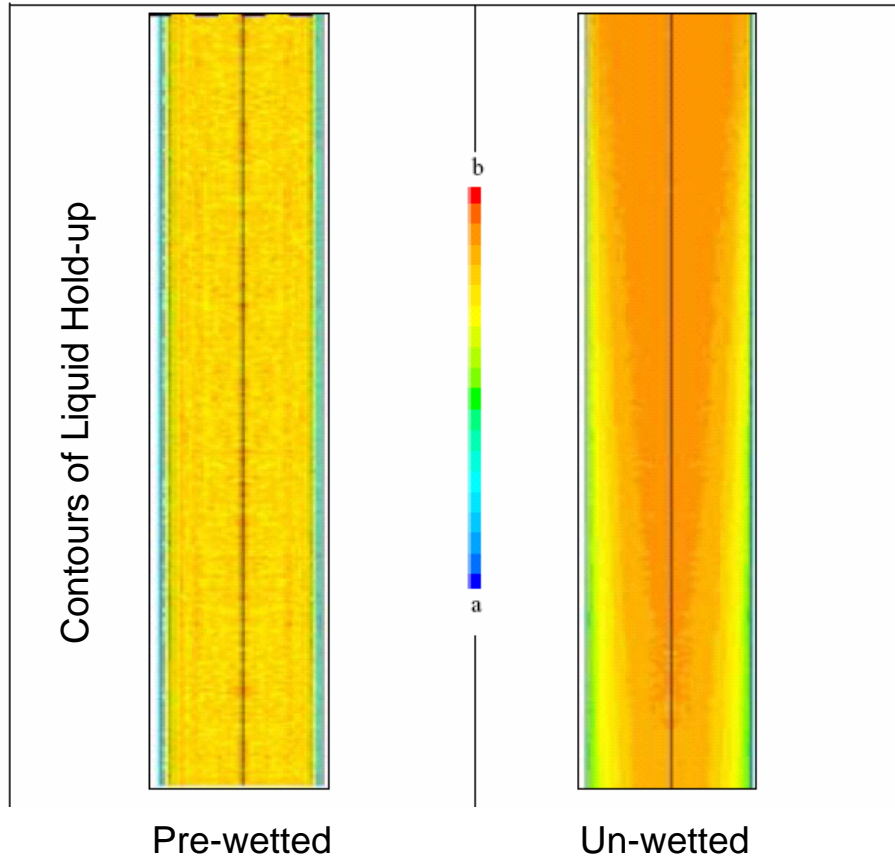
$$F\left(\frac{\rho_G}{\rho_L}\right) = 1 + 88.1 \frac{\rho_G}{\rho_L} \quad \text{for } \frac{\rho_G}{\rho_L} < 0.025$$

– Extent of Wetting, f (Jiang et al., 2002)

$$P_G - P_L = (1 - f)P_c$$

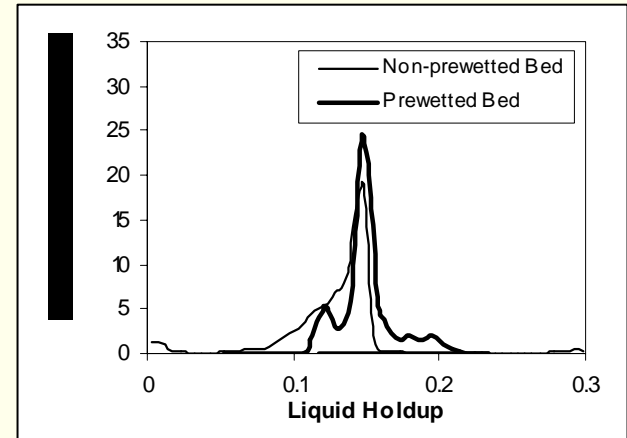
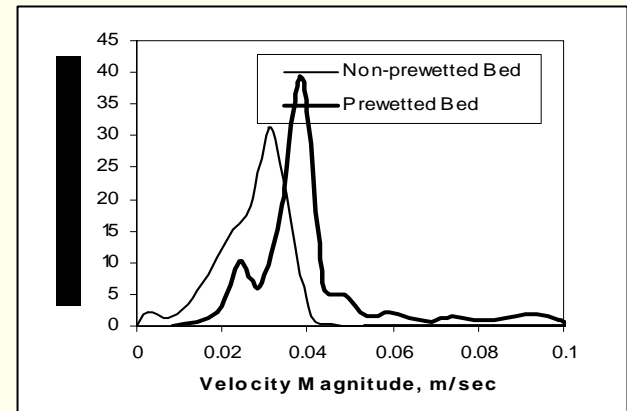
- Scalar Transport

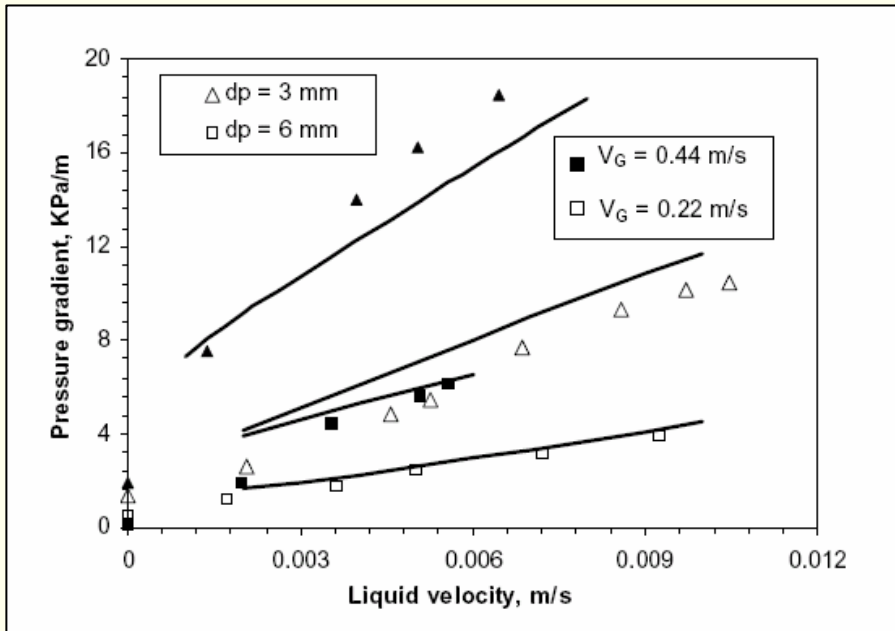
$$\frac{\partial \varepsilon_K \rho_K C_i}{\partial t} + \nabla \cdot \varepsilon_K \rho_K U_K C_i = -\nabla \cdot (\varepsilon_K \rho_K D_{i,m} \nabla C_i) + S_i$$



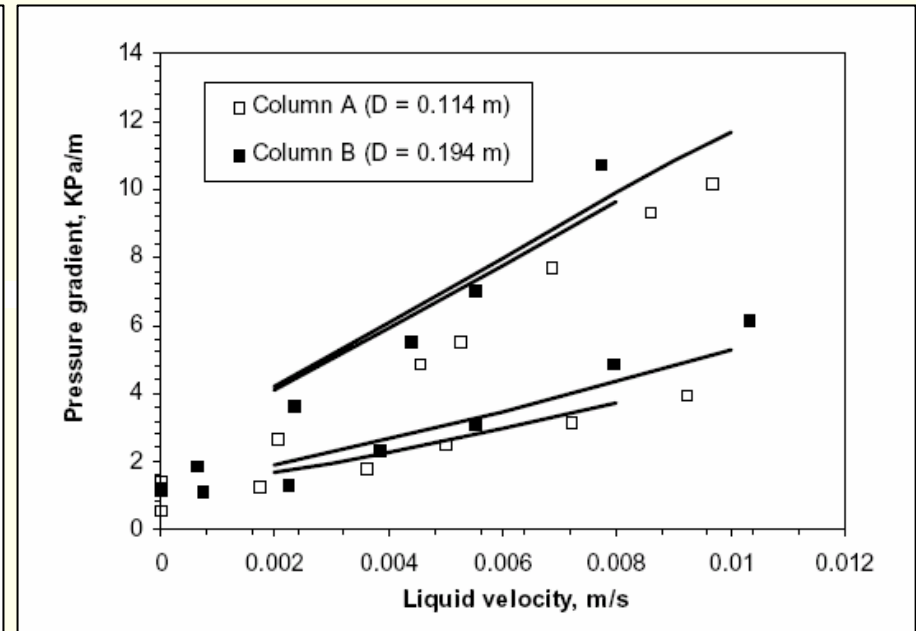
$V_L = 6 \text{ mm/s}$, $V_G = 0.22 \text{ m/s}$, $D = 0.114 \text{ m}$, $d_p = 3 \text{ mm}$, $\sigma = 5\%$

Distributions within the bed

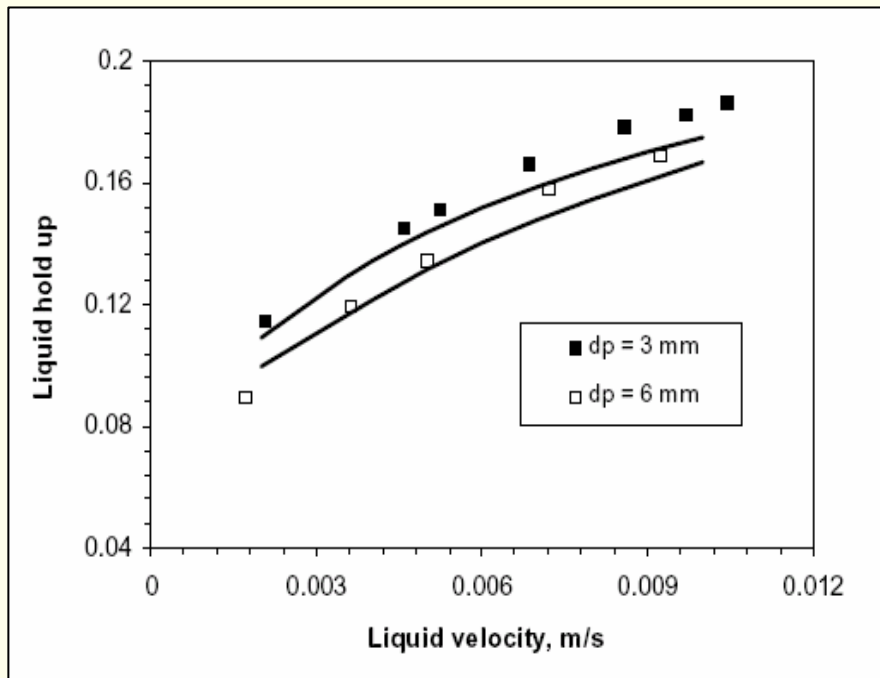




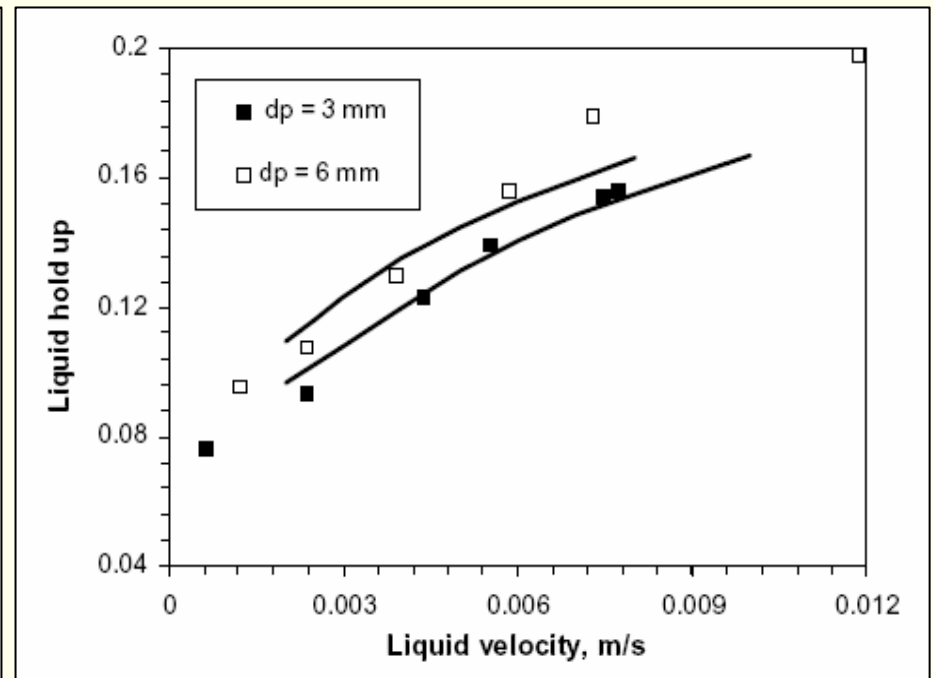
Column Diameter = 0.114 m



Particle Diameter = 3 mm, V_G = 0.22 m/s



$V_G = 0.22$ m/s, $D = 0.114$ m



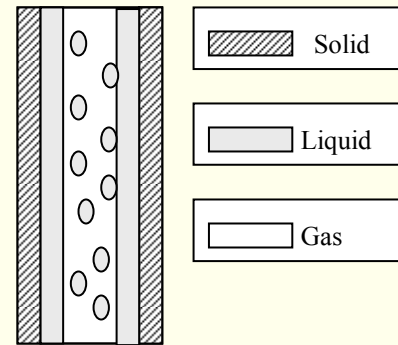
$V_G = 0.22$ m/s, $D = 0.194$ m



- Estimation of Frictional Pressure Drop & Fraction of Liquid Supported by Gas

$$\left(\frac{\Delta P}{L}\right)_{GL} = \left(\frac{\Delta P_f}{L}\right)_{GL} - (\rho_L \phi_L g)$$

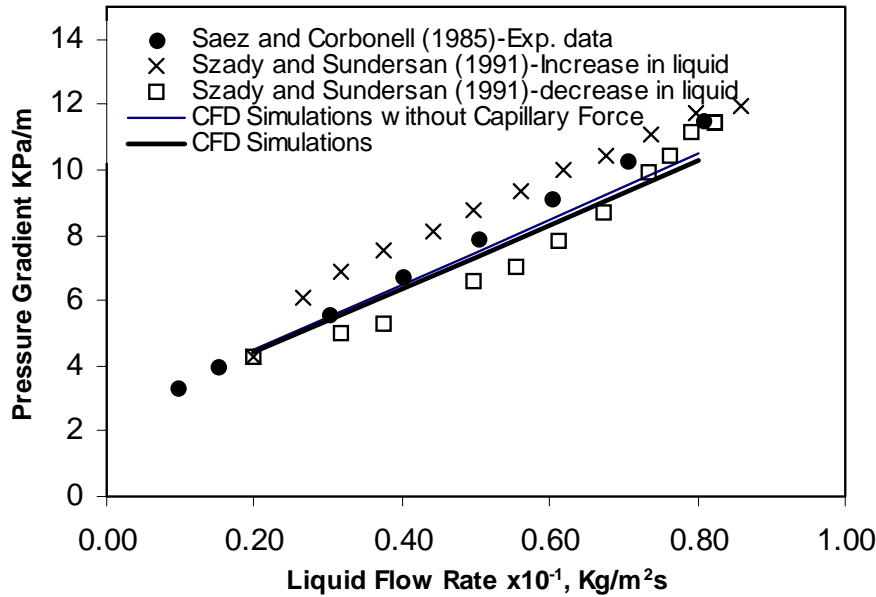
$$\therefore \phi_L \leq \beta_L$$



Fraction of Liquid Hold-up Supported by Gas Phase < 1

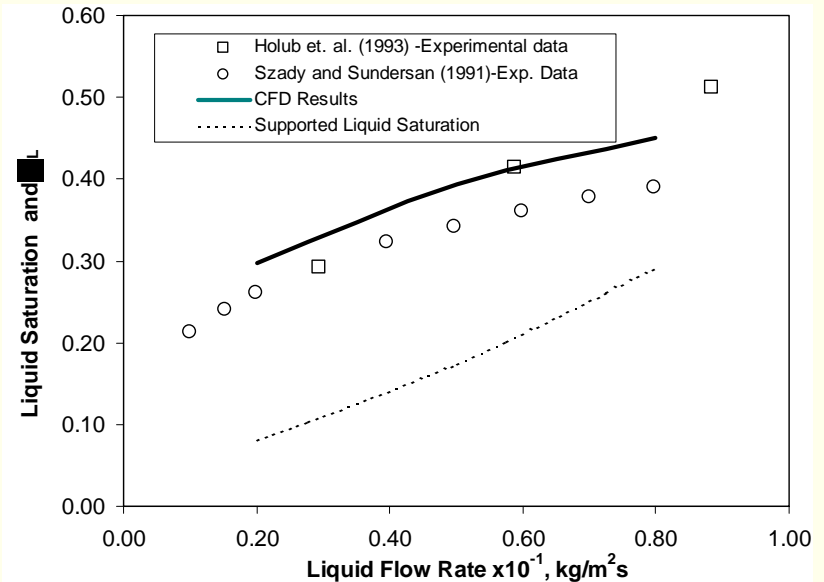
- Simulate at two values of 'g' (9.7, 9.9 m/s²)

$$\left(\frac{\Delta P_f}{L}\right) = \frac{g_1 \left(\frac{\Delta P}{L}\right)_2 - g_2 \left(\frac{\Delta P}{L}\right)_1}{g_1 - g_2} \quad \phi_L = \frac{\left(\frac{\Delta P}{L}\right)_2 - \left(\frac{\Delta P}{L}\right)_1}{\rho_L (g_1 - g_2)}$$

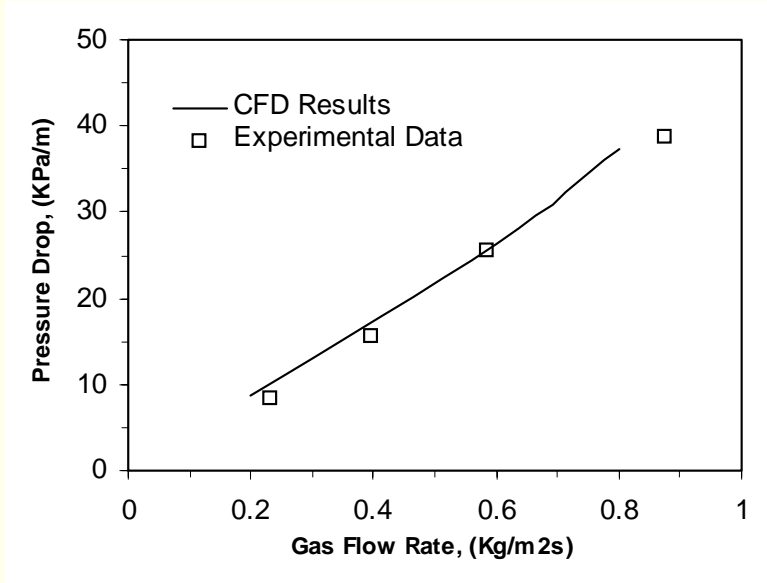


Can Predict Total and Supported Liquid Volume Fraction

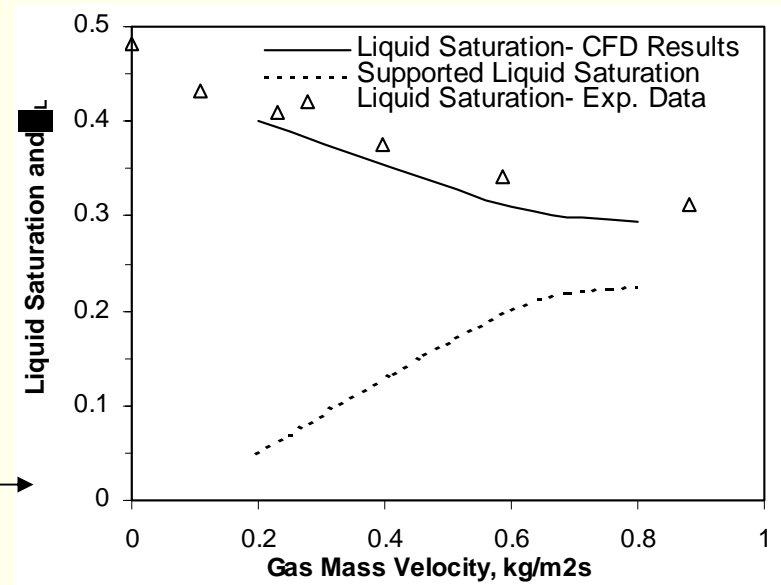
Calibrate Model Parameters from ΔP



Operating conditions: $V_G=0.22\text{m/s}$, $D/d_p=55$, $d_p=3\text{mm}$, Std. Dev.=5%, $E_1=215$, $E_2=1.75$

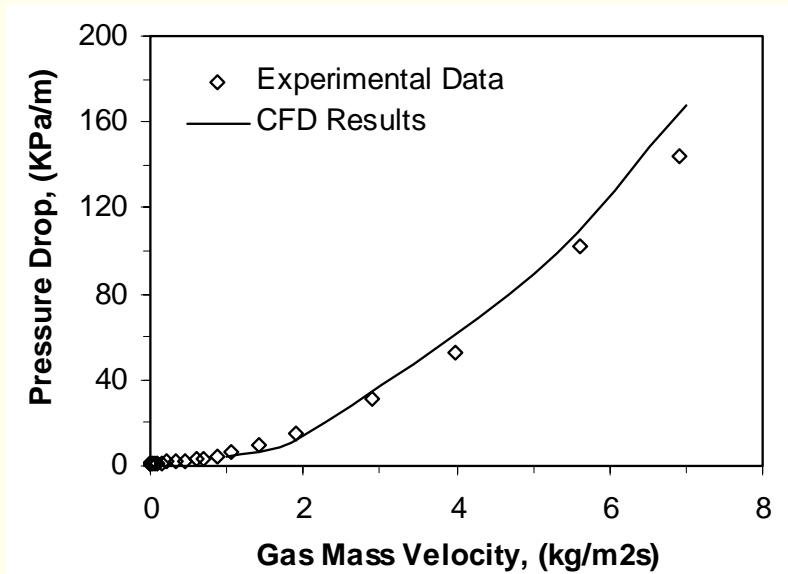


Data of Spachio & Baldi (1977)

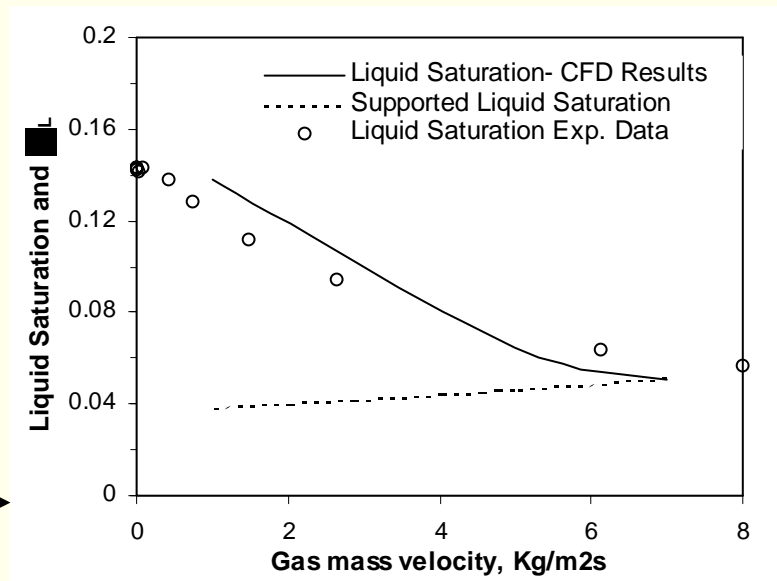


As gas velocity increases,
more & more liquid is supported by gas

Operating conditions: $V_L=2.8 \times 10^{-3}$ m/s, $d_p=3$ mm, $D/d_p=30$, Std. Dev.=5%, $E_1=500$, $E_2=3$

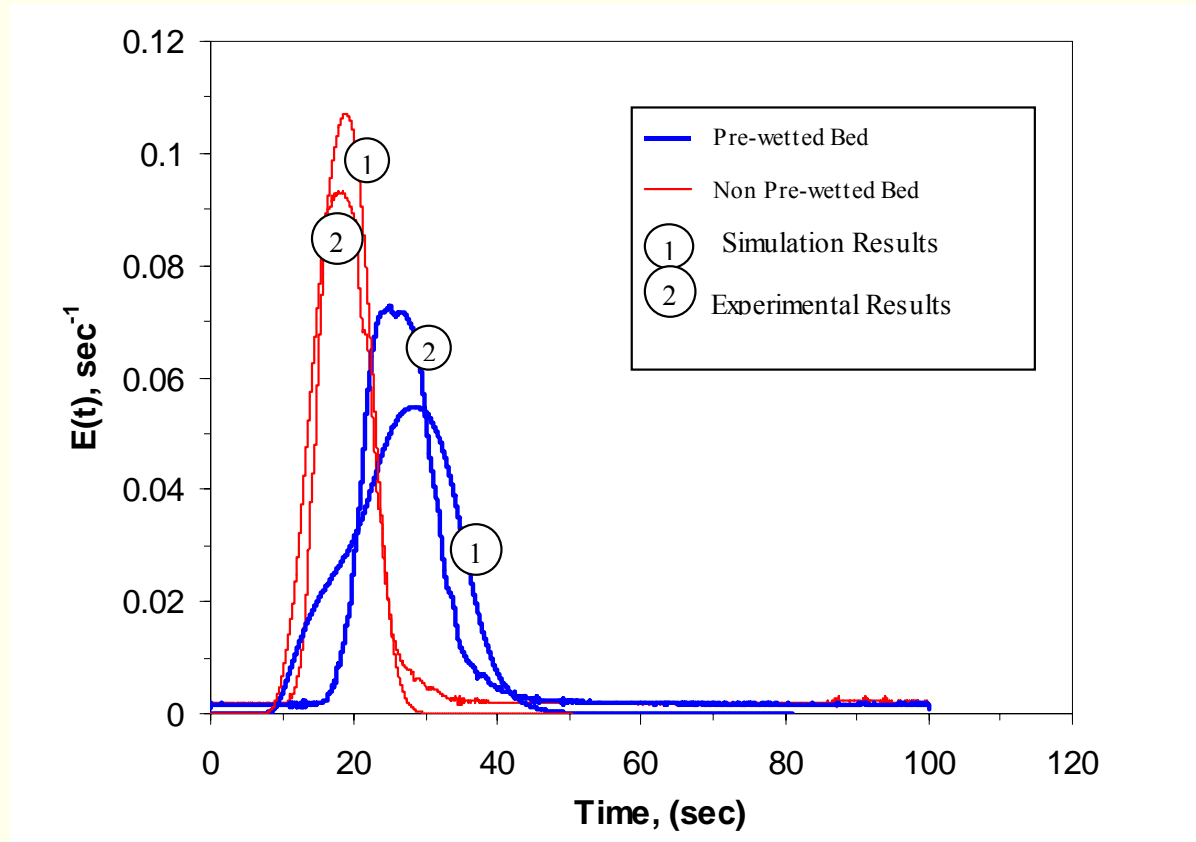


Data of Rao et al. (1983)



As gas velocity increases,
 more & more liquid is
 supported by gas

Operating conditions: $V_L = 1 \times 10^{-3} \text{ m/s}$, $D/d_p = 15.4$, $d_p = 3 \text{ mm}$, Std. Dev. = 5%, $E_1 = 215$, $E_2 = 3.4$

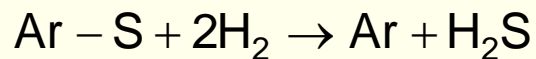


Operating conditions: $V_L=2.06 \times 10^{-3} \text{m/s}$ $V_G=0.22 \text{m/s}$, $D/d_p=18$, $d_p=6 \text{mm}$, Std. Dev.=5%, $E_1=180$, $E_2=1.75$



- Reaction and Kinetics (Chowdhury et al. 2002)

Hydro-desulfurisation (HDS)



$$r_{\text{HDS}} = -\frac{k c_{\text{L,H}_2}^{0.56} c_{\text{L,S}}^{1.6}}{1 + K_{\text{Ad}} c_{\text{L,H}_2\text{S}}}$$

De-aromatisation of Mono-, Di- and Poly- Aromatics





- Pressure drop is insignificant compared to the operating pressure
- Trickle bed reactor is operated isothermally (efficient heat transfer)
- Ideal gas law is applicable
- Liquid phase reactants are non-volatile (negligible vapor pressure)
- Gas-liquid mass transfer is the limiting resistance.
- The catalyst particles are completely wetted



- Mass Balance of Species i

$$\frac{\partial \varepsilon_k \rho_k C_{k,i}}{\partial t} + \nabla \cdot (\varepsilon_k \rho_k U_k C_{k,i}) = -(\varepsilon_k \rho_k D_{i,m} \nabla C_{k,i}) + \varepsilon_k \rho_k S_{i,k}$$

- Source Terms for Gas Phase

$$S_i = -K_{GLi} a_{GL} \left[\frac{C_{Gi}}{H_i} - C_{Li} \right]$$

- Source Terms for Liquid Phase

$$S_i = K_{GLi} a_{GL} \left[\frac{C_{Gi}}{H_i} - C_{Li} \right] + \rho_B \eta \sum_{j=1}^{j=nr} r_{ij}$$

$$S_i = \rho_B \eta \sum_{j=1}^{j=nr} r_{ij}$$



- Mass Transfer Coefficient (From Goto & Smith, 1975)

$$\frac{k_i^L a_L}{D_i^L} = \alpha_1 \left(\frac{G_L}{\mu_L} \right)^{\alpha_2} \left(\frac{\mu_L}{\rho_L D_i^L} \right)^{1/2}$$

- Solubility of Hydrogen/ H₂S (Korsten et al. 1996)

$$H_i = \frac{v_N}{\lambda_i \cdot \rho_L} \quad v_N : \text{molar gas volume}$$

$$\lambda_{H_2} = a_0 + a_1 T + a_2 \frac{T}{\rho_{20}} + a_3 T^2 + a_4 \cdot \frac{1}{\rho_{20}^2}$$

$$\lambda_{H_2S} = \exp(3.3670 - 0.008470 \cdot T)$$

$$a_0 = -0.559729$$

$$a_1 = -0.42947 \times 10^{-3}$$

$$a_2 = 3.07539 \times 10^{-3}$$

$$a_3 = 1.94593 \times 10^{-6}$$

$$a_4 = 0.835783$$



- Gas and Liquid Superficial Velocities Increase with Scale
- Wetting gets Better with Scale

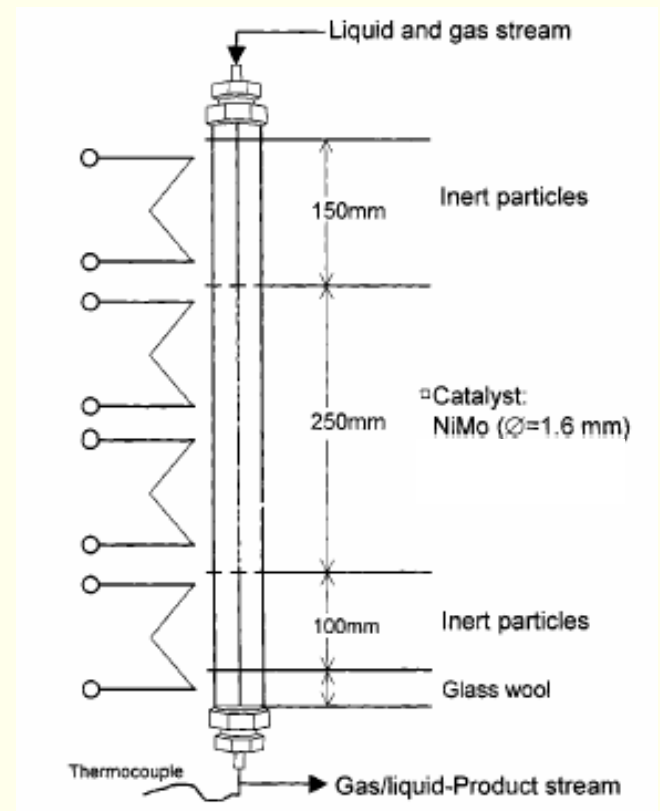
Parameters	Laboratory Scale Reactor (Chowdhury et al. 2002)	Commercial Scale Reactor (Bhaskar et al. 2004)
Reactor Diameter, m	0.019	3.8
Bed Length	0.5 m	16 m
Particle Diameter, m	0.0024	0.0015
Bed Porosity	0.50	0.36
LHSV, h-1	1-5	1-5
Operating Pressure, Mpa	20-28	20-80
Operating Temperature, K	573-693	573-693
Initial H ₂ S Conc., v/v %	0.5-8	0.5-8

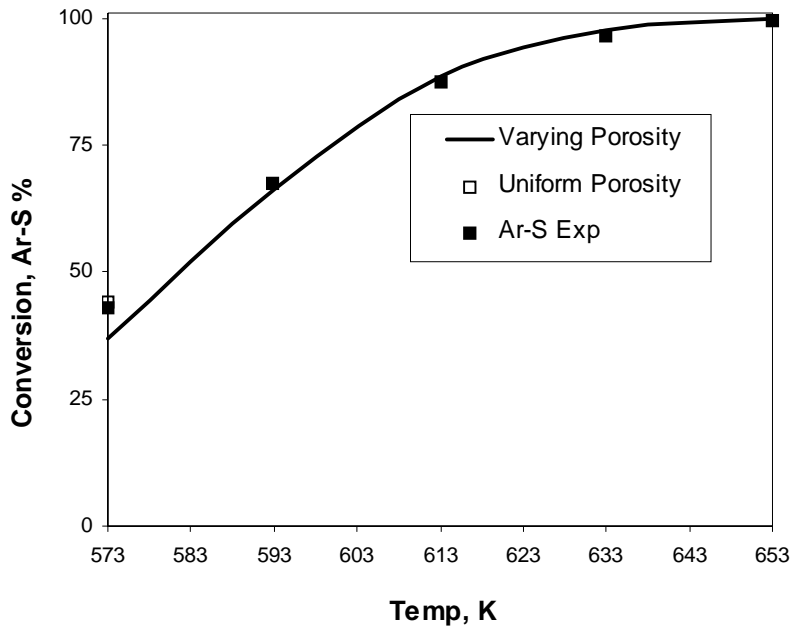


- From Chowdhury et al. (2002)

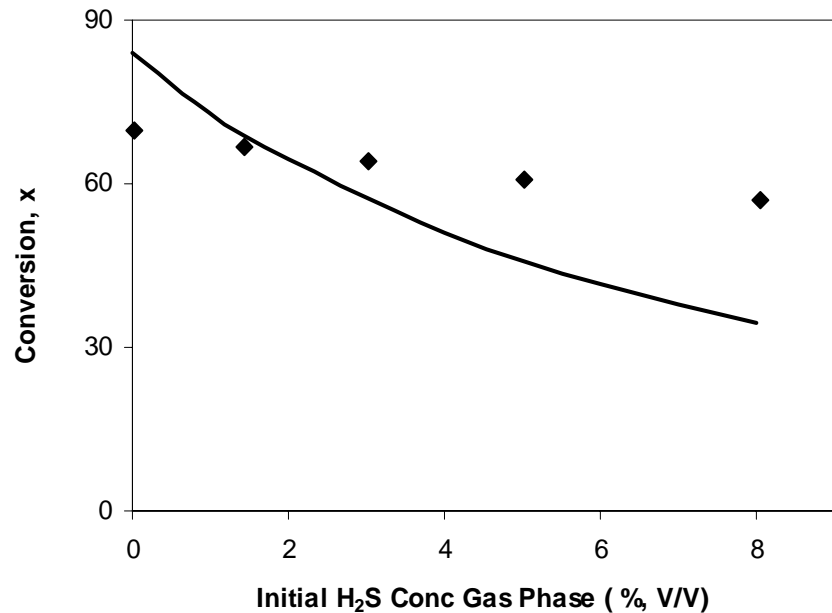
<u>Kinetic Constants</u>	<u>Values</u>
K_{Ad} , $m^3/kmol$	50000
K, Dimensionless	$2.5 \times 10^{12} \exp(-19384/T)$
k^*_{mono} , $m^3/kg.s$	$6.04 \times 10^2 \exp(-12414/T)$
k^*_{Di} , $m^3/kg.s$	$8.5 \times 10^2 \exp(-12140/T)$
k^*_{poly} , $m^3/kg.s$	$2.66 \times 10^5 \exp(-15170/T)$

<u>Component</u>	<u>Percentage</u>
Ar-S %	1.67
Poly-Ar %	2.59
Di-Ar %	8.77
Mono-Ar %	17.96
Naphthenes %	19.25



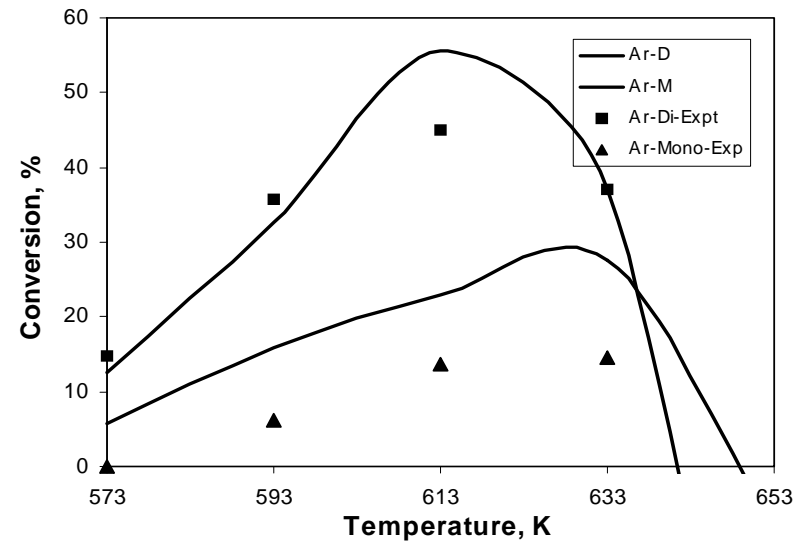
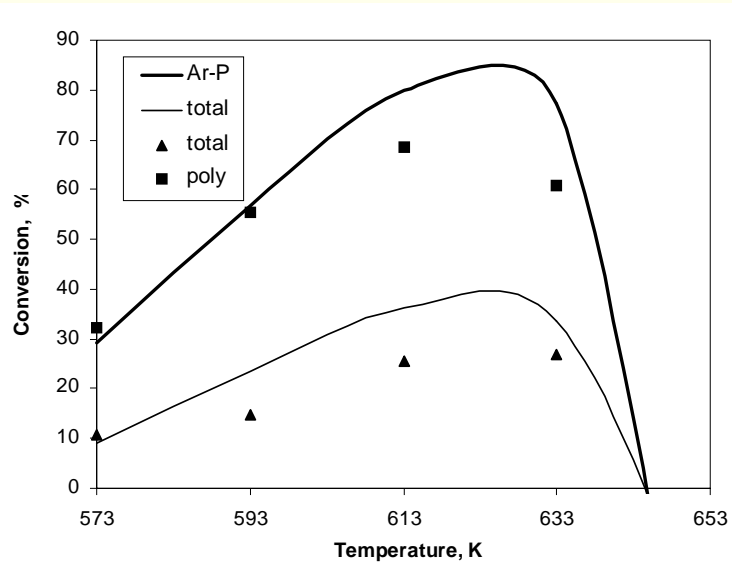


Temperature

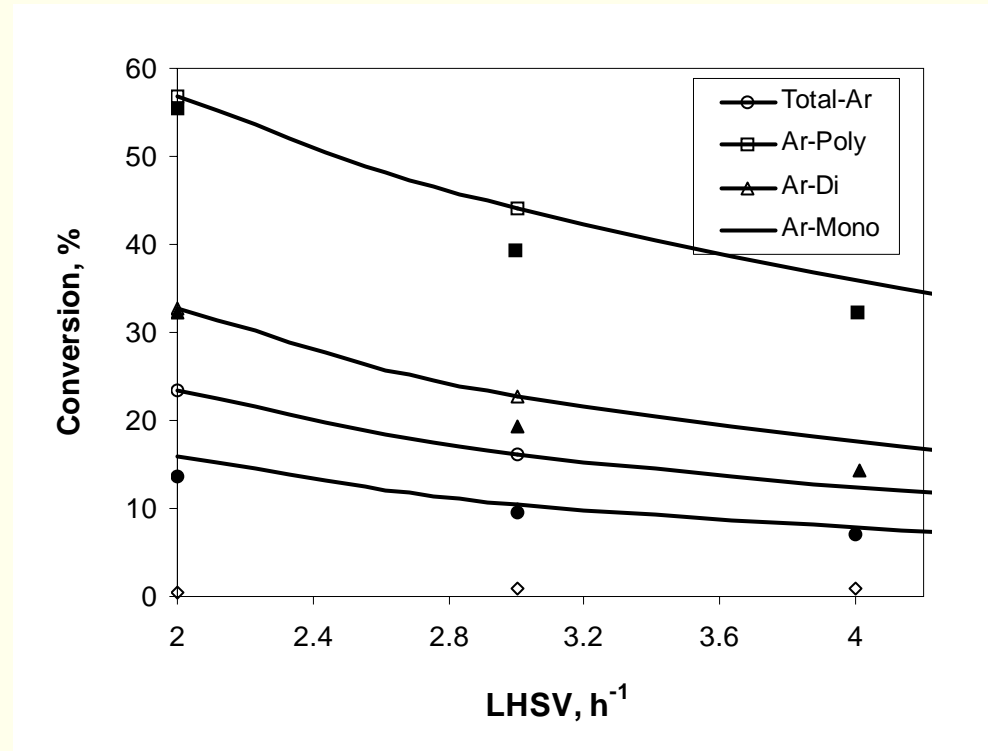


Initial Concentration of H₂S

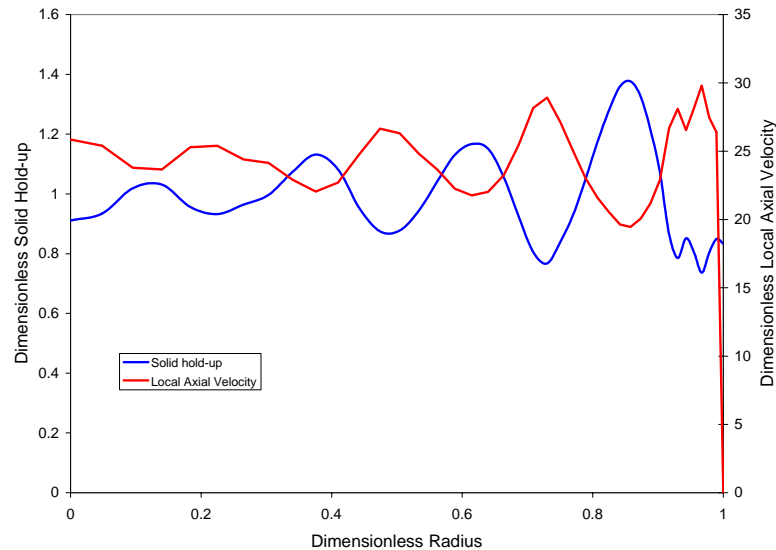
($P=4$ MPa, $LHSV=2.0$ h⁻¹, $Q_{GNTP}/Q_L=200$ m³/m³, $T_R=320^\circ\text{C}$, $y_{H_2S}=1.4\%$)
 Symbols denote experimental data of Chowdhury et al. (2002)



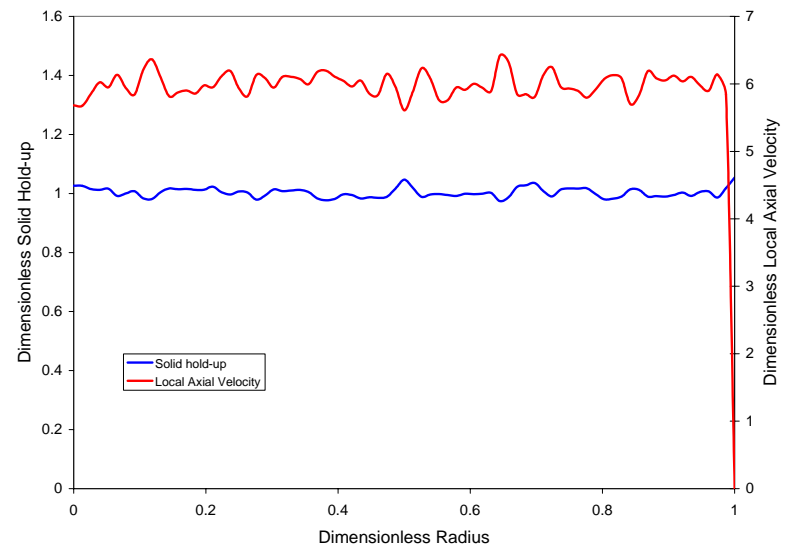
($P=4$ MPa, $LHSV=2.0$ h^{-1} , $Q_{GNTP}/Q_L=200$ m^3/m^3 , $y_{H_2S}=1.4\%$)
Symbols denote experimental data of Chowdhury et al. (2002)



($P=4$ MPa, $TR=320$ °C, $Q_{GNTF}/Q_L=200$ m³/m³, $y_{H_2S}=1.4\%$, filled symbols are for experimental data of Chowdhury et al., 2002)

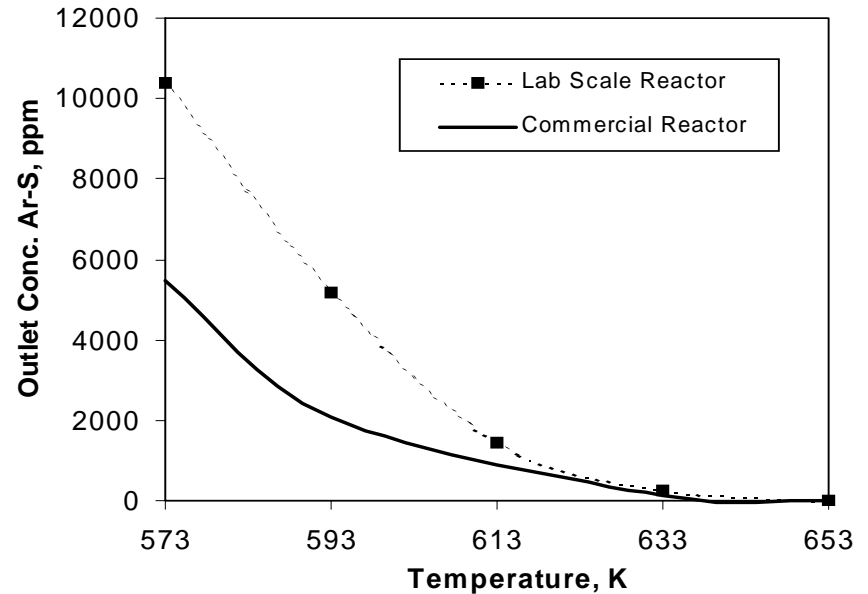
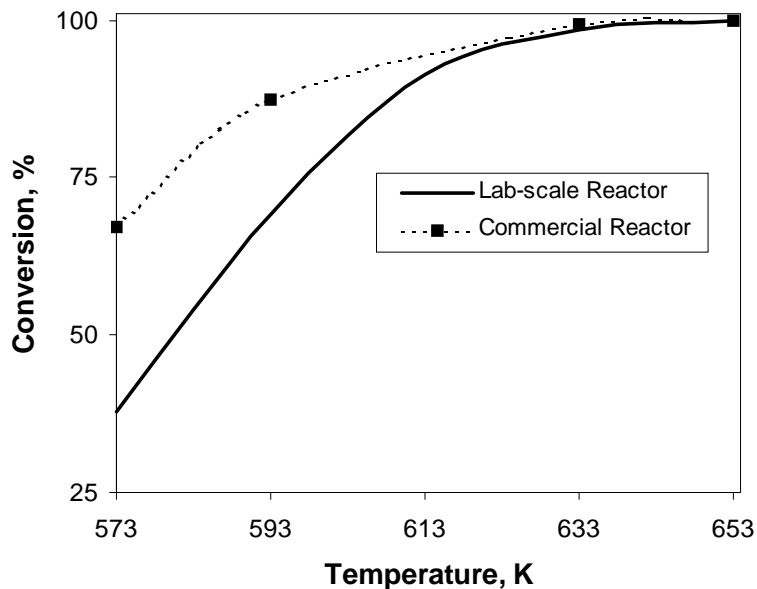


Laboratory (LHSV=3)

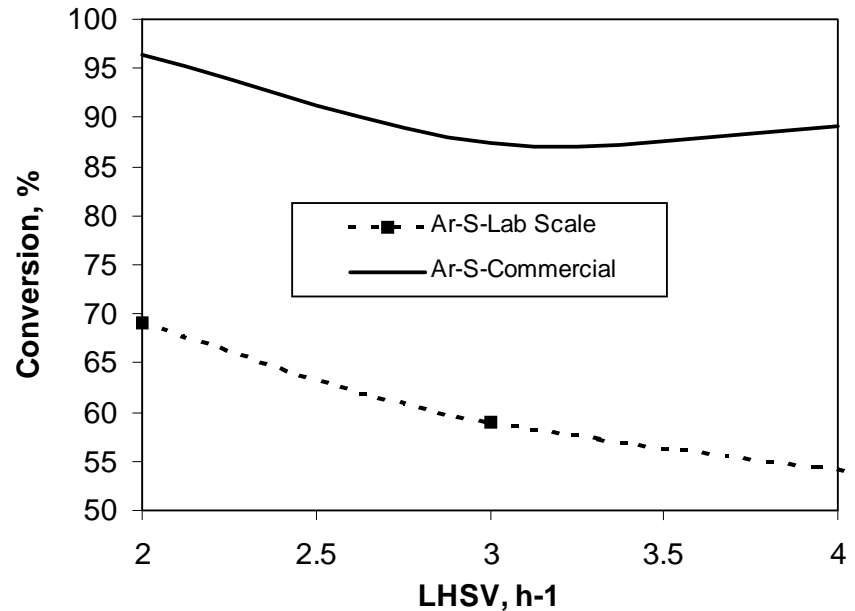
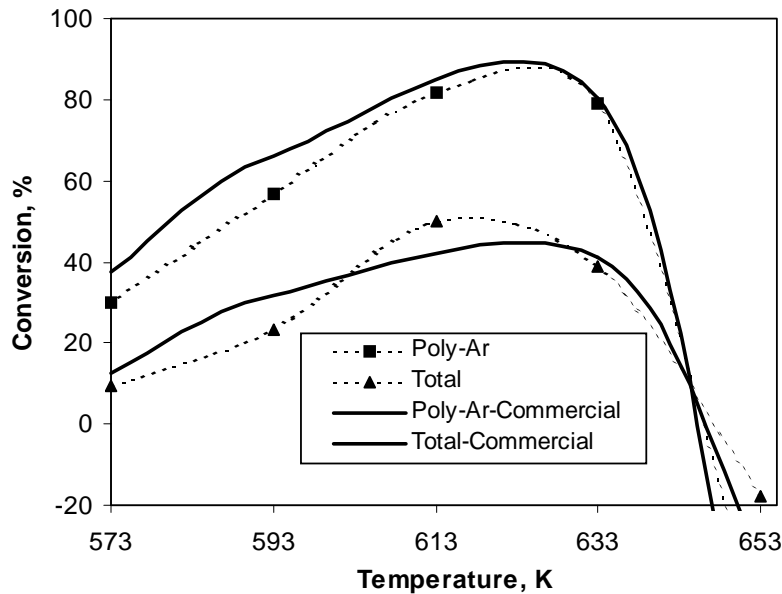


Industrial (LHSV=2)

($P=4$ MPa, $T_R=320$ °C, $Q_{GNTP}/Q_L=300$ m³/m³)

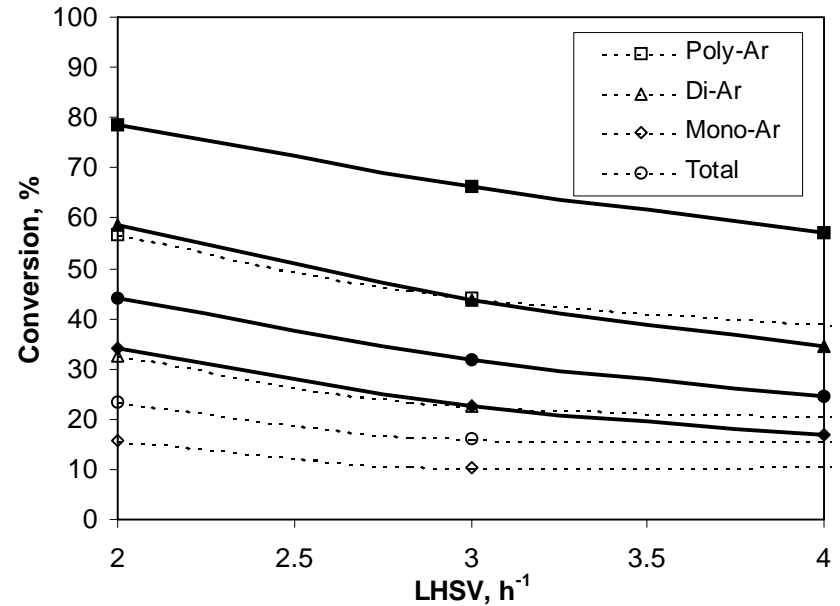
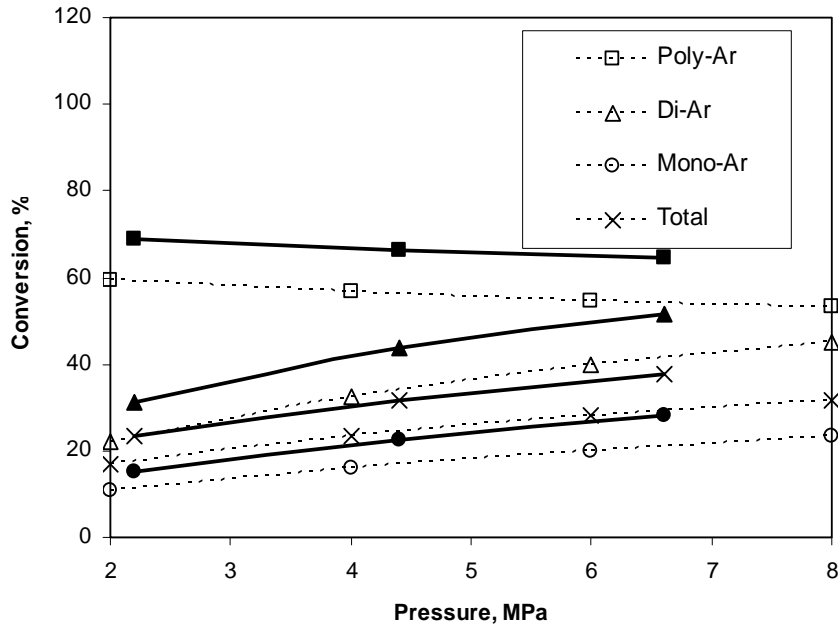


($P_{\text{lab \& com}} = 4 \text{ \& } 4.4 \text{ MPa}$, $LHSV_{\text{lab}} = 2.0 \text{ h}^{-1}$, $Q_{\text{GNTP}}/Q_{\text{L}} = 200 \text{ m}^3/\text{m}^3$, $y_{\text{H}_2\text{S}} = 1.4\%$)



($P_{lab \& com} = 4 \& 4.4 \text{ MPa}$, $LHSV_{lab \& com} = 2.0 \& 3.0 \text{ h}^{-1}$,
 $(Q_{GNTP}/Q_L)_{lab \& com} = 200 \& 300 \text{ m}^3/\text{m}^3$, $y_{H_2S} = 1.4\%$)

($P_{lab \& com} = 4 \& 4.4 \text{ MPa}$, $Temperature = 593 \text{ K}$,
 $(Q_{GNTP}/Q_L)_{lab \& com} = 200 \& 300 \text{ m}^3/\text{m}^3$, $y_{H_2S} = 1.4\%$)



$(Q_{GNTP}/Q_L)_{lab \& com} = 200 \& 300 \text{ m}^3/\text{m}^3$ $T=593 \text{ K}$
 $LHSV_{lab \& com} = 2.0 \& 3.0 \text{ h}^{-1}$, $y_{H_2S} = 1.4\%$

$P_{lab \& com} = 4 \& 4.4 \text{ Mpa}$, $T=593 \text{ K}$ $y_{H_2S} = 1.4\%$
 $(Q_{GNTP}/Q_L)_{lab \& com} = 200 \& 300 \text{ m}^3/\text{m}^3$

Continuous lines: commercial; Dotted lines: laboratory scale



- Macroscopic CFD Model Reasonably Simulates Gas-liquid Flow in Trickle Beds
 - Liquid hold-up
 - Residence time distribution
 - May be used to estimate fraction of suspended liquid
- Further Work is Needed for Understanding Wetting/Hysteresis to Make Further Progress
- Despite Limitations, CFD Models can be Used to Understand Key Issues in Reactor Engineering Including Scale-up & Scale-down

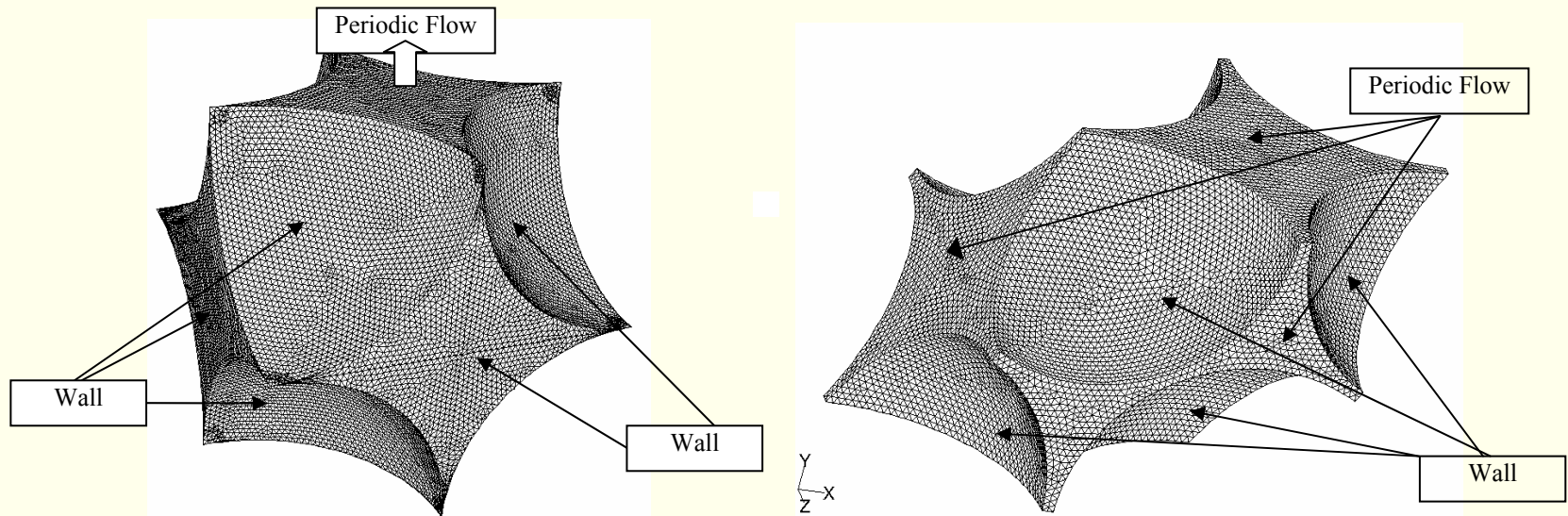




- **Single Phase Flow through Packed Bed**
 - Unit cell approach
 - Simple cubic, FCC, rhombohedral ..
 - Inertial flow structures, pressure drop, heat transfer
- **Interaction of Liquid Drop/ Film with Solid Surface**
 - Regimes of interaction
 - Dynamic contact angle/ surface characteristics
 - VOF simulations
 - Insight into capillary forces???



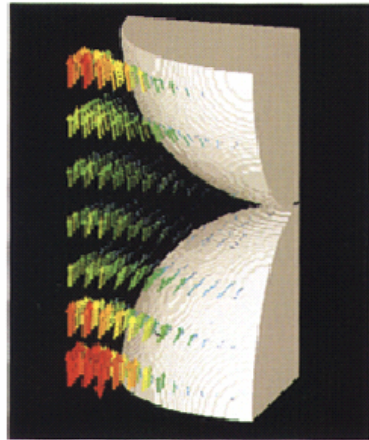
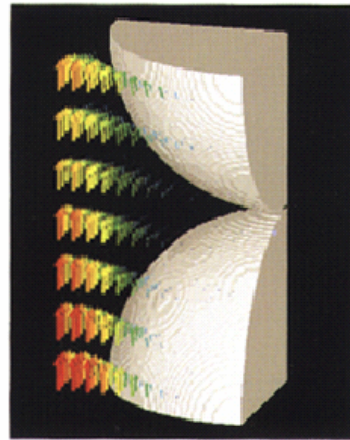
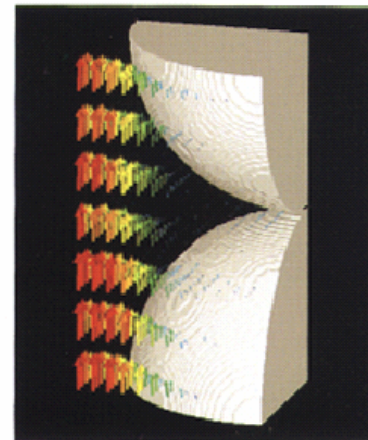
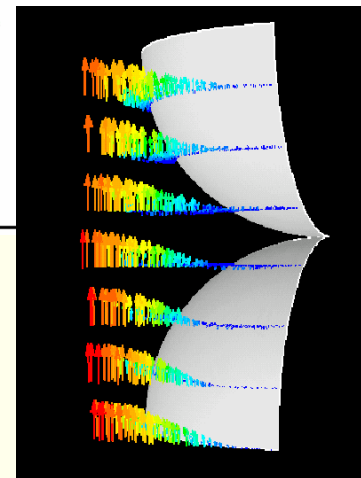
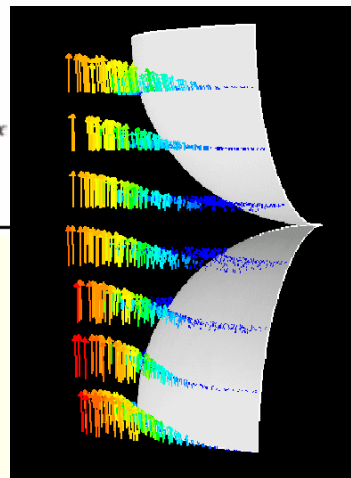
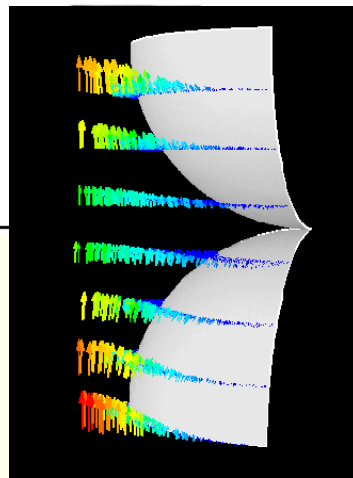
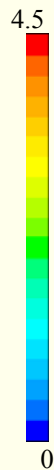
- Single Phase Flow through Packed Bed



Operating Parameters

- Cell Length 28mm, 3mm
- Fluid Water
- Particle Reynolds Number 12-6000

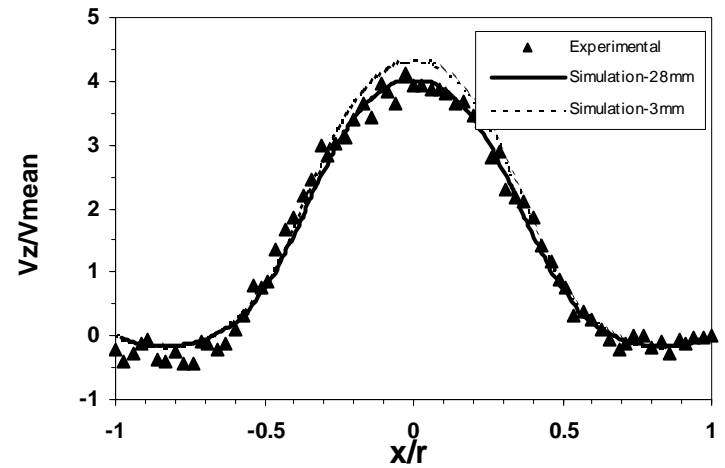
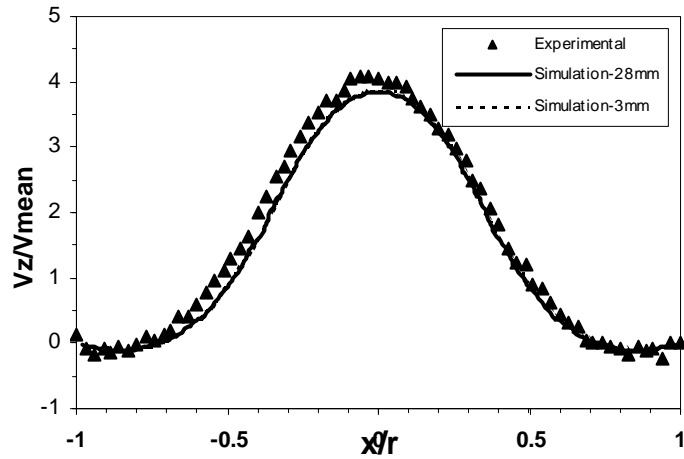
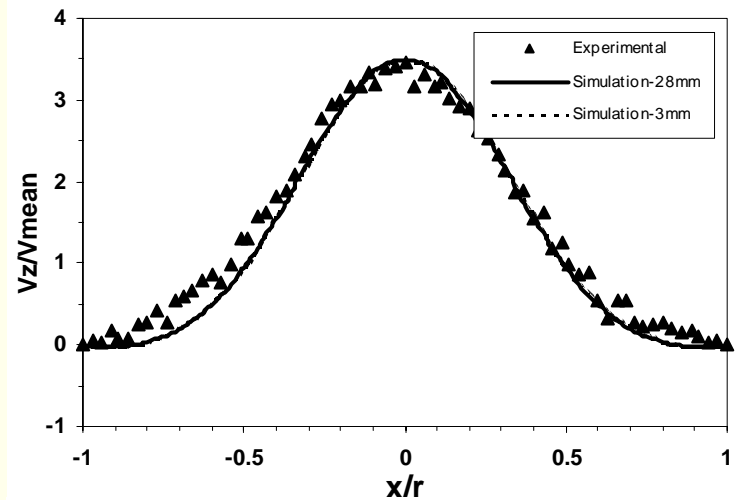
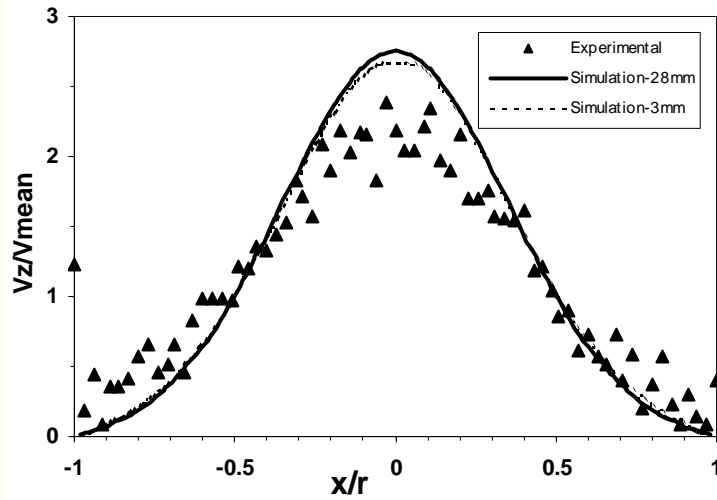
SINGLE PHASE FLOW

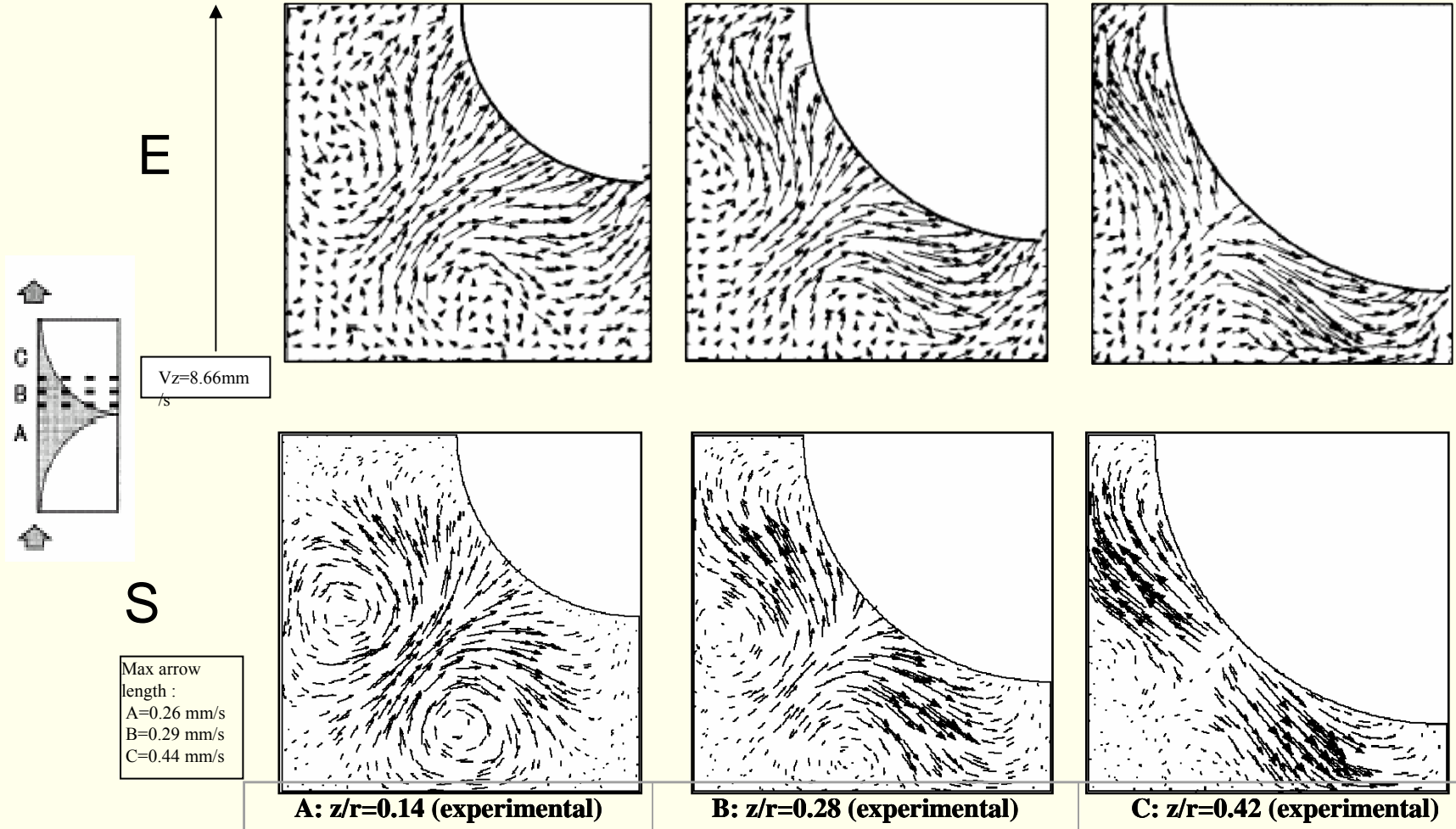
(a) $Re=12.17$ (b) $Re=59.78$ (c) $Re=204.74$ 

Experiments:

Suekane et al.
AIChE J., **49**, 1

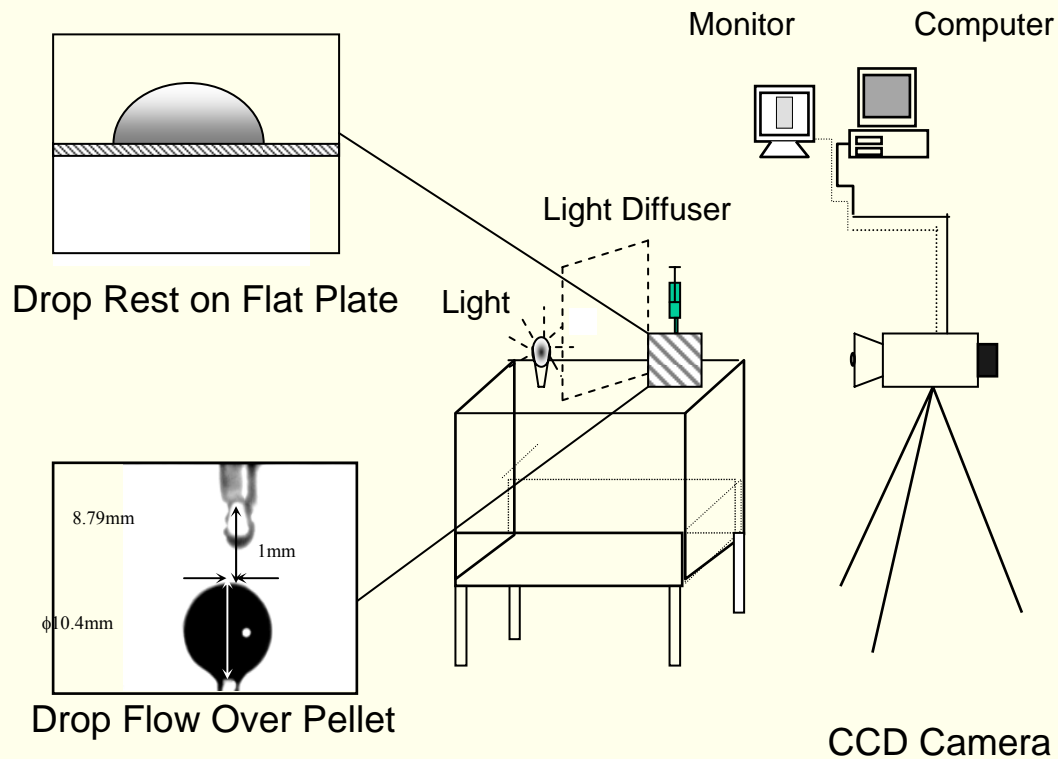
Simulations

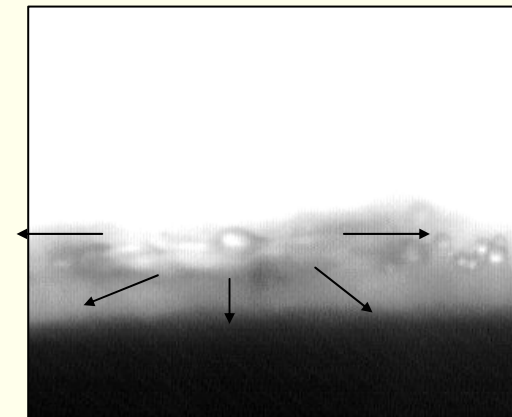
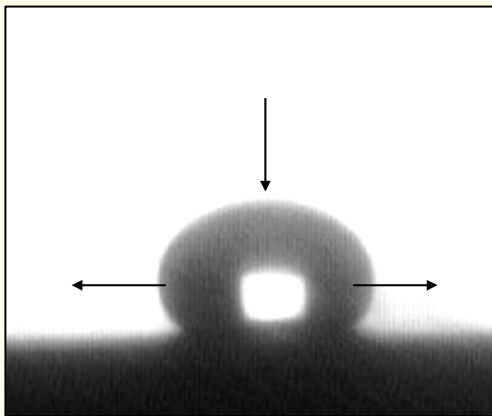
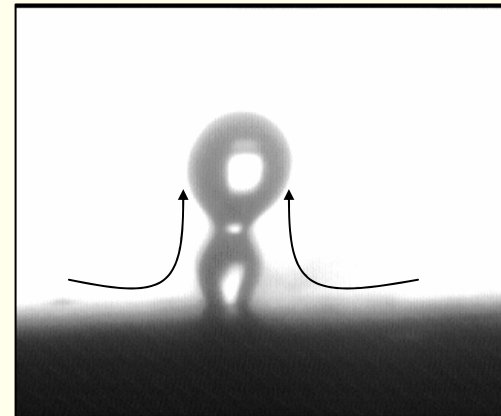
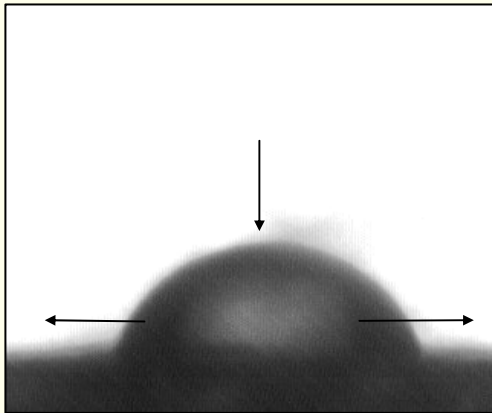


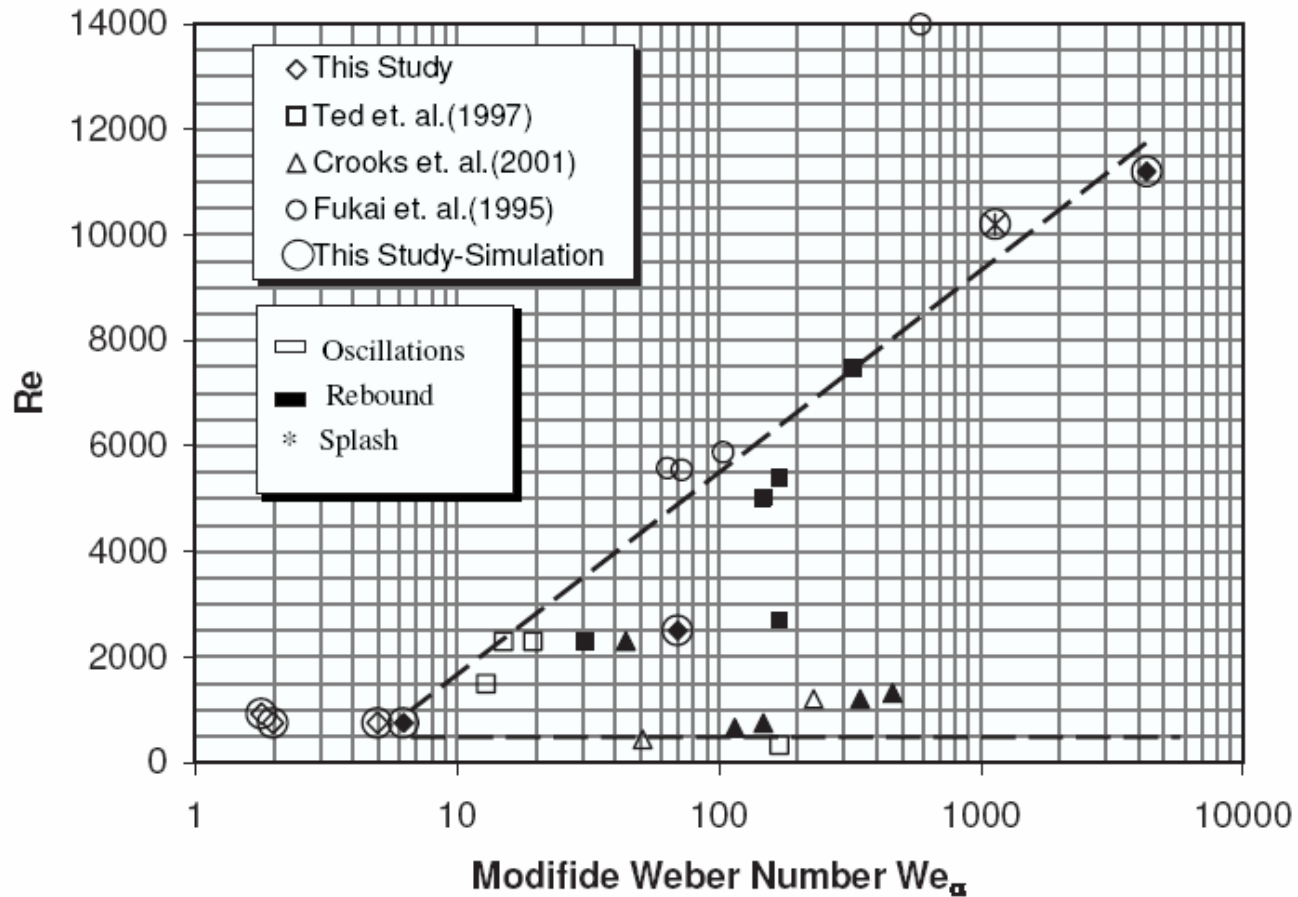


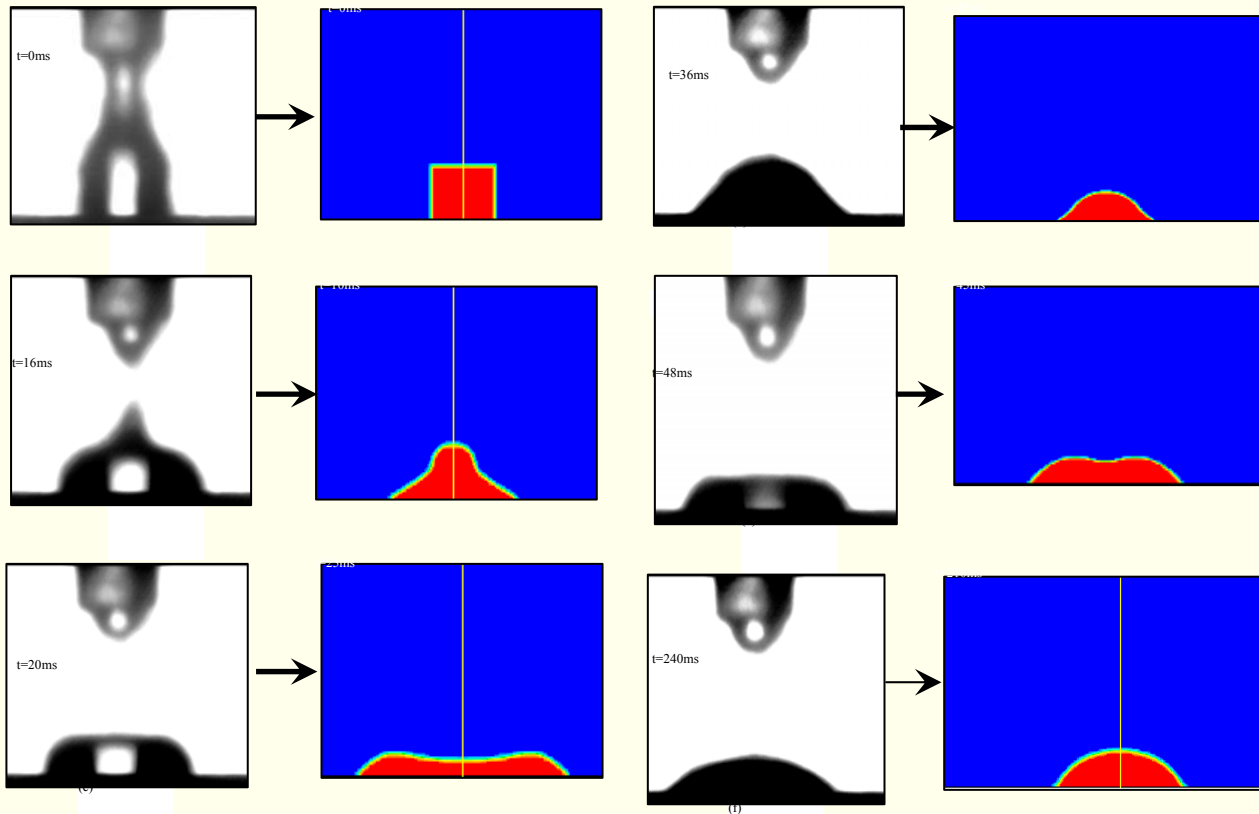


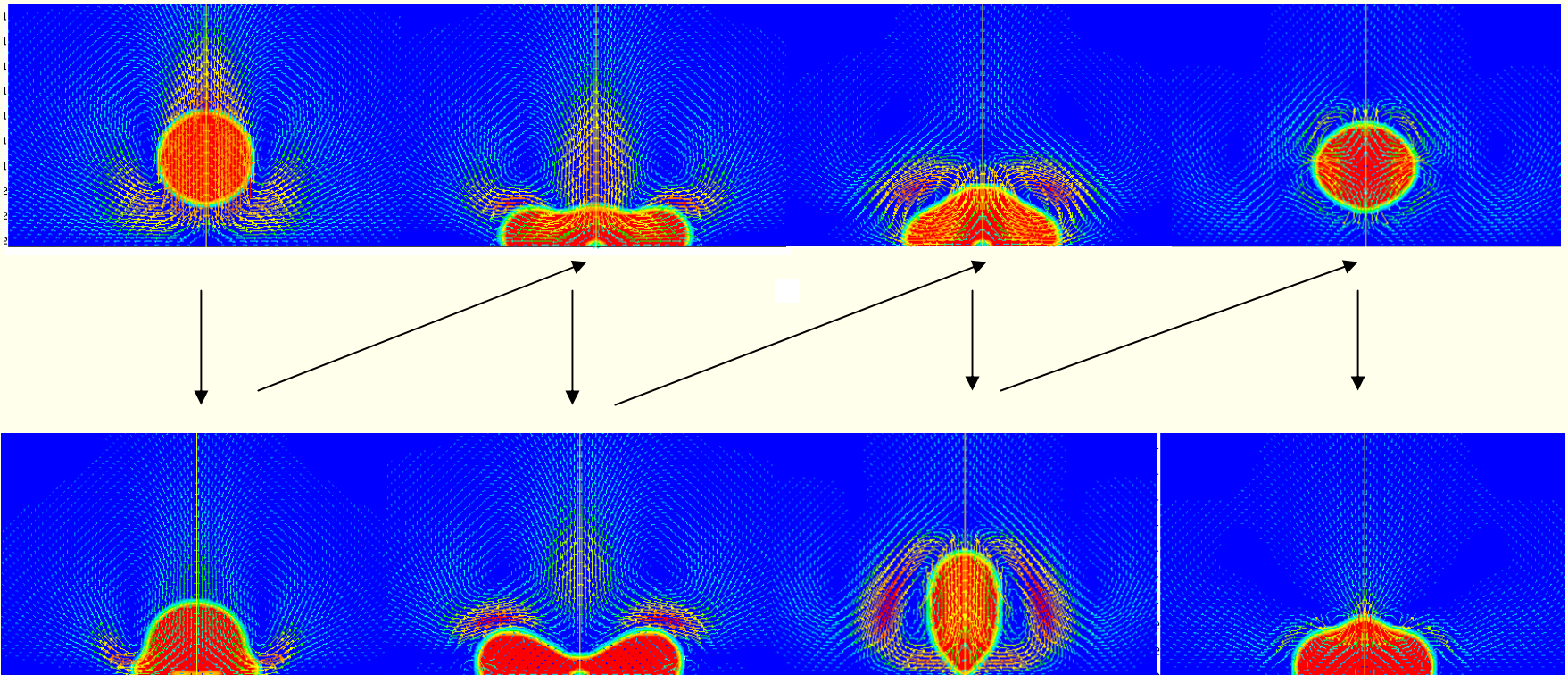
- Inertial Flow Structures were Captured Accurately by CFD Models
- Velocity Distribution in Unit Cells Resembles that in a Packed Bed
- Ergun Parameters may be Obtained through Unit Cell Approach for Semi-structured Packed Bed
- Unit Cell Approach may be Extended to Simulate Two Phase Flows to Understand Wetting

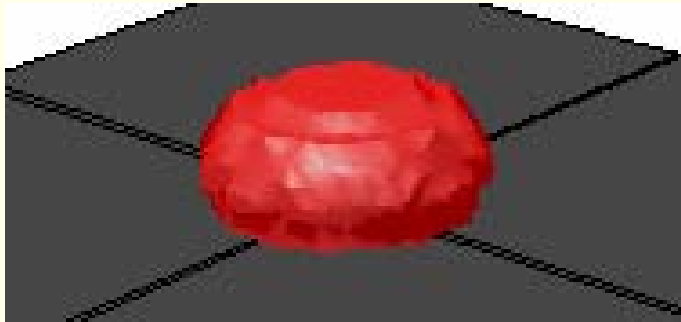




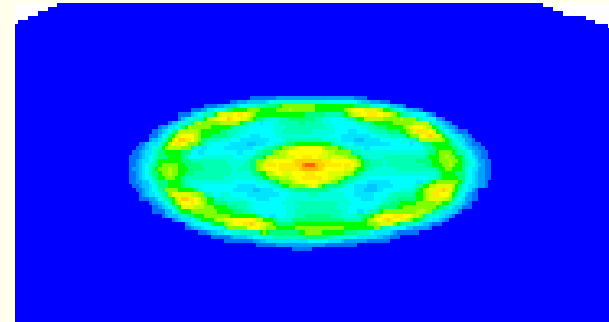








Drop Shape at 12 ms



Wall Shear Stress on
Solid Surface, red~100 Pa

Shear
Strain on
Gas-liquid
Interface

